Thermal Folding and Mechanical Unfolding Pathways of Protein Secondary Structures

Marek Cieplak,^{1,2,*} Trinh Xuan Hoang,³ and Mark O. Robbins¹

¹Department of Physics and Astronomy, The Johns Hopkins University, Baltimore, Maryland ²Institute of Physics, Polish Academy of Sciences, Warsaw, Poland ³International School for Advanced Studies (SISSA), Trieste, Italy

ABSTRACT Mechanical stretching of secondary structures is studied through molecular dynamics simulations of a Go-like model. Force versus displacement curves are studied as a function of the stiffness and velocity of the pulling device. The succession of stretching events, as measured by the order in which contacts are ruptured, is compared to the sequencing of events during thermal folding and unfolding. Opposite cross-correlations are found for an α -helix and a β -hairpin structure. In a tandem of two α -helices, the two constituent helices unravel nearly simultaneously. A simple condition for simultaneous versus sequential unraveling of repeat units is presented. Proteins 2002;49:104-113. © 2002 Wiley-Liss, Inc.

Key words: mechanical stretching of proteins; protein folding; Go model; molecular dynamics; atomic force microscopy; α-helix; β-hairpin

INTRODUCTION

Weak noncovalent bonding forces govern functioning and structural cohesion in cells. Direct measurements of these forces through mechanical means has recently become an important tool in studies of biological molecules. There is a variety of techniques for probing forces in the pico- and nano-Newton range,¹ such as atomic force microscopy,²⁻⁵ optical tweezers,^{6,7} the surface force apparatus,⁸ micropipette aspiration,⁹ and the quartz microbalance.^{10,11} As examples of recent achievements, we list elucidation of the nature of interactions of a chaperone protein (HIV-1) with DNA through stretching of a strand of the DNA with optical tweezers¹² and discovery of stick-slip motion when two strands of a DNA double helix are pulled apart.¹³

The techniques used in mechanical unfolding of individual biological molecules rely on tethering of the molecule between movable surfaces. This tethering is relatively easy to accomplish with long molecules such as DNA and giant proteins such as titin,^{6,7} which are naturally built as a tandem array of many globular domains. For shorter molecules, the pulling surfaces interact and affect the pulled molecule in a way that makes the data hard to interpret. In order to extend the method to single domained proteins, Yang et al.¹⁴ have recently developed a method of synthesizing identical repeats of protein molecules in the solid state, which were then studied using a modified scanning force microscope. This technique has been applied to T4 lysozyme.

At this moment, experimental data on the mechanical unfolding of the secondary structures of proteins are not available. However, data on periodically repeated proteins and even individual proteins may become available in the near future. From a theoretical point of view, it is important to gain an understanding of the basic unfolding mechanisms of simple structures and to develop analytical tools that could then be used for large proteins. This process is facilitated by considering simple models that allow a rapid exploration of parameter space. Our choice in this article is to analyze Go-like models,¹⁵ which emphasize the importance of native conformations and treat non-native interactions only schematically. The Go-like models,¹⁵ though coarse-grained, are fairly realistic¹⁶ in their kinetic properties and allow for a thorough characterization and comparison of mechanical, equilibrium, and folding properties in a straightforward manner. This kind of full characterization is difficult to achieve in all-atom models with the Amber¹⁷ or CHARMM¹⁸ force fields. These models are perhaps best suited to studies of mechanical stretching, but even there are restricted to rapid stretching rates because of their high computational cost.

The idea that mechanical unfolding experiments on proteins have the potential to provide insights into the relevant folding pathways is what motivated Bryant et al.¹⁹ to carry out all-atom (CHARMM-based) simulations of the C-terminal hairpin of protein G, the folding of which has been previously studied experimentally by Munoz et al.^{20,21} They have found that, under low pulling forces, breakdown of hydrogen bonds precedes dissociation of the hydrophobic cluster. Their interpretation of this finding is that thermal folding should proceed in the opposite order to mechanical unfolding. If so, then the zippering folding mechanism²¹ would be less favored than one in which a

Grant sponsor: National Science Foundation; Grant number: DMR-0083286. Grant sponsor: Theoretical Interdisciplinary Physics and Astrophysics Center (Johns Hopkins). Grant sponsor: KBN (Poland); Grant number: 2P03B-146-18.

^{*}Correspondence to: Marek Cieplak, Institute of Physics, Polish Academy of Sciences, Al. Lotnikow 32-46 02-668 Warsaw, Poland. E-mail: mc@ifpan.edu.pl

Received 16 October 2001; Accepted 2 May 2002

hydrophobic cluster is formed first. This prediction remains to be tested.

Key differences between folding and unfolding have also been emphasized. One of the most important is that mechanical forces break the isotropy of the system. Socci and coworkers²² have identified transitions between different unfolding regimes as the magnitude of the force increases. At small forces the unfolding time is dominated by thermal mechanisms. Intermediate forces produce a weakly anisotropic environment that biases thermally activated breaking of bonds that are oriented along the force. Still larger forces produce a highly anisotropic state, and the protein unravels rapidly without any need for thermal activation.

Here, we explore properties of Go models of proteins through molecular dynamics simulations. We consider the variant in which contact interactions are described by Lennard-Jones potentials. The simulations include a Langevin noise term that both mimics presence of a solvent and controls the temperature, *T*. This article focuses on an α -helix of 16 monomers, denoted as H16; a β -hairpin of 16 monomers, B16; and a double repeat of the α -helix, H16-2. The companion article²³ describes a similar analysis for titin.

We first study mechanical unfolding at nearly zero temperature. This choice of T minimizes fluctuations and rate dependence and most simply reveals the effects of the structure of the energy landscape. The results should be equivalent⁵ to fast stretching at higher T, and temperature dependence will be considered in subsequent work. The protein is stretched by a Hookean cantilever, and the force is plotted as a function of the cantilever displacement. We characterize the stretching process by studying the succession of unfolding events, which are described by the cantilever displacements at which specific contacts are broken. Both the force-displacement curve and the order of unfolding events depend on the stiffness and velocity of the cantilever. The soft spring limit corresponds to the constant force case of Socci et al.,²² whereas the stiffer springs are comparable to those used in all atom simulations (e.g., Ref. 19).

We next discuss studies of folding, where temperature plays an essential role. The sequencing of folding events depends on T, and smooth and simple pathways are only found near an optimal temperature denoted by T_{\min} .^{24–28} The sequencing of folding events near T_{\min} is contrasted with that of stretching events for different protein structures. We find that both sequencings are governed primarily by the contact order, 29-31 i.e., by the distance between two amino acids along the sequence of the protein. However, the cross-correlations between thermal and mechanical sequencings are opposite for the two simple cases considered: H16 and B16. Only in the latter case do folding and stretching occur in the opposite order, as envisioned by, e.g., Bryant et al.¹⁹ In general, the thermal and mechanical pathways can be very different and their relation depends critically on the way force is distributed among bonds.

Another quantity that we study here is what we propose to call an irreversibility length, $L_{\rm ir}$. If one studies folding from a fully extended conformation, then one finds that the characteristic folding time diverges as $T \rightarrow 0$. Thus, a fully stretched protein will not fold back to the native state at low temperatures. On the other hand, a protein that is pulled only slightly will return to its native shape on release. There must then be a characteristic stretched length of the protein, $L_{\rm ir}$, which separates the two behaviors. We demonstrate that $L_{\rm ir}$ does indeed exist and find that it is substantially larger for B16 than for H16. Furthermore, the folding time for lengths less than $L_{\rm ir}$ is a complicated function of the mechanical extension.

We also consider a tandem arrangement of two α -helices and find that the constituent helices unravel almost simultaneously whereas in titin²³ the unraveling is serial in nature. Simple criteria for the two types of behavior are described.

MODEL AND METHOD

The model we use is described in detail in References 26–28 and 32. For simplicity, we consider the variant where steric constraints associated with dihedral and other angles are ignored. Briefly, a protein is modeled by a chain of identical beads that correspond to the locations of the C^{α} atoms. The consecutive beads interact through the potential³³

$$V^{BB} = \sum_{i=1}^{N-1} [k_1(r_{i,i+1} - d_0)^2 + k_2(r_{i,i+1} - d_0)^4], \quad (1)$$

where $\mathbf{r}_{i,i+1} = |\mathbf{r}_i - \mathbf{r}_{i+1}|$ is the distance between two consecutive beads, $d_0 = 3.8$ Å is the equilibrium bond length, $k_1 = \epsilon/\text{Å}^2$, $k_2 = 100 \epsilon/\text{Å}^4$, and ϵ is the characteristic energy parameter corresponding to a native contact. The anharmonic term in Equation (1) prevents energy localization in specific phonons and thus accelerates equilibration.³³

The interaction that governs the native contacts (defined as those C^{α} that are not immediate neighbors, but are no further than 7.5 Å apart in the native structure) is chosen to be of the Lennard-Jones type (see e.g., Ref. 34):

$$V^{NAT} = \sum_{i < j}^{NAT} 4\varepsilon \left[\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^{6} \right].$$
(2)

The parameters σ_{ij} are chosen so that each contact in the native structure is stabilized at the minimum of the potential, and $\sigma \equiv 5\text{\AA}$ is a typical value. As a technical criterion for determining when a native contact forms or breaks during the time evolution, we adopted the cutoff value of $1.5\sigma_{ij}$. The non-native contacts are described by purely repulsive potentials. These are obtained by evaluating V^{NAT} with a length parameter σ , truncating the potential at its minimum $(2^{1/6}\sigma)$, and shifting it to have zero value at this cutoff distance.

Figure 1 illustrates the forms of the potentials for the α -helix. When studying the folding times, we have adopted





Fig. 2. Snapshot pictures of stretching of the α -helix H16. The left end is anchored elastically, and the right end is pulled by the stiff cantilever. The numbers indicate the cantilever displacements. In the top panel the helix is still almost fully folded. Note that the native bonds between turns of the helix are aligned with the pulling force.

Fig. 1. The potentials used to construct a Go model of the α -helix H16. The interactions are between the beads *i* and *i*+*k*. For *k* = 1 this is the anharmonic tethering potential. The contact corresponding to *k* = 5 is non-native and is thus purely repulsive. The remaining contact interactions are of the Lennard-Jones form.

a simplified approach in which a protein is considered folded if all beads that form a native contact are within the cutoff distance of $1.5\sigma_{ij}$ instead of making a more precise delineation of the native basin as in Reference 27. This will allow for a more meaningful comparison with the results on titin.²³

The beads are coupled to Langevin noise and damping terms to mimic the effect of the surrounding solution and maintain constant temperature T. The equations of motion for each bead are

$$m\ddot{\mathbf{r}} = -\gamma\dot{\mathbf{r}} + F_c + \Gamma, \qquad (3)$$

where m is the mass of the amino acids represented by each bead, F_c is the net force due to the molecular potentials and external forces, γ is the damping constant, and Γ is a Gaussian noise term with dispersion $\sqrt{2\gamma k_{\rm B}T}$. We measure time in units of the characteristic period of undamped oscillations in the Lennard-Jones potential $\tau \equiv \sqrt{m\sigma^2/\epsilon}$. Using typical values for the average amino acid mass, length, and binding energy yields 3ps as an estimate of τ . According to Veitshans et al.,³⁵ realistic estimates of damping by the solution correspond to a value of γ near 50 m/τ . However, the folding times have been found to depend on γ in a simple linear fashion for $\gamma >$ m/τ .^{26,27,36} Thus, in order to accelerate the simulations, we work with $\gamma = 2m/\tau$. The equations of motion are solved by means of the fifth-order Gear predictor-corrector algorithm³⁷ with a time step of 0.005τ .

In order to pull the protein apart, we attach both of its ends to purely harmonic springs of spring constant *k*. We focus on three cases: (a) the stiff spring: $k \ge 60 \epsilon/\text{Å}^2$, (b) the

soft spring: $k = 0.12 \epsilon/Å^2$, and (c) the very soft spring: $k = 0.04\epsilon/Å^2$. The outer end of one spring is held stationary, and the other is pulled at a fixed rate v_p . This model's stretching by a Hookean cantilever with stiffness k/2, because the two springs add in series. We also performed simulations at constant force, which corresponds to the limit of infinitely weak springs. However, the unwinding of the proteins occurs in an "all or nothing" fashion in this limit, and little information can be extracted.

The pulling direction is chosen to coincide with the initial end-to-end vector of the protein. In general, the molecule reorients rapidly to maximize the end-to-end length along the pulling force. This reorientation was studied for titin by Lu and Schulten.³⁸ In most cases, we pull the spring very slowly—at a constant rate of v_p = $0.005\text{\AA}/\tau$. There is actually very little dependence of the results on pulling rate until one considers rapid rates. For instance, increasing v_p by a factor of 50 produces almost no change in the force. Substantial rate dependence begins when v_p is increased by a factor of 100 to 0.5Å/ τ , and this case is denoted as a "fast" stretch in the following section. The instantaneous pulling force F is the extension of the pulling spring times the spring constant k. Plotted values of F are averaged over 1τ . The standard pulling velocity is low enough that the force equilibrates along the chain and almost the same force is obtained from the extension of the spring whose end is fixed. Drag terms lead to a significant difference in these forces at higher velocities. The force is plotted versus the cantilever displacement $d = v_n t$, where *t* is the total pulling time.

RESULTS AND DISCUSSION

α-Helix

Figure 2 illustrates the process of mechanical unfolding for H16. It clearly shows that unfolding starts at both ends



Fig. 3. Force versus displacement for H16 with stiff and soft cantilevers. Solid lines, slow pulling rates; dotted lines, fast pulling rates.

and then proceeds to the center. This is precisely the ordering of events during thermal folding²⁶ and *not* the inverse of this ordering as seen for the C-terminal hairpin of protein G.¹⁹ However, the underlying reasons for the observed ordering during folding and unfolding are different. Folding starts at the ends because they diffuse more rapidly and are thus more likely to fall into a contact situation, whereas unfolding starts at the ends because there are fewer binding forces there. Note that the axis of the helix is roughly parallel to the pulling force during unfolding. Bonds along the entire length of the helix are nearly aligned with the force and all are placed under comparable tensile stresses. This is very different than the β -hairpin case discussed below.

The force versus cantilever displacement curves are shown in Figure 3 for slow (solid lines) and fast (dashed lines) displacement rates. The curves are truncated when the helix is fully stretched, and any further displacement results in rapid growth in F followed by rupture of the protein backbone. The top and bottom panels are for the stiff and soft pulling springs, respectively. In both cases unfolding produces a sequence of stick-slip events. The force rises linearly while the protein is trapped in a given local energy minimum and then drops rapidly as one or more contacts break. The slope of the upward rise is the combined stiffness k_{tot} of the protein k_p and the cantilever k/2. Because the two are in series, $k_{tot}^{-1} = k_p^{-1} + 2k^{-1}$. In the soft spring case the cantilever dominates, and the slope of the upward ramps is k/2. For the stiff cantilever case the internal stiffness of the protein dominates. Variations within and between local minima lead to changes in the slope of the ramps, with k_p varying between ~ 0.3 and 0.6

 $\epsilon / {\rm \AA}^2$. Once k is larger than these values, it has little influence on the curves.

Each upward ramp ends when one or more contacts break. The force drops sharply until the protein reaches a new metastable state and a new upward ramp begins. In the low velocity case (solid lines), \boldsymbol{v}_p is much lower than the velocities produced by contact breaking, and rupture occurs at a nearly constant cantilever position. In this limit, the force drop is roughly equal to $k_{\rm tot}$ times the change in protein length during the jump between metastable states. For a stiff cantilever (top panel), the failure of each contact produces a large drop in the force. The first two peaks correspond to breaking of the two end contacts. The force is lower than for later events because the ends have fewer native contacts. Rupturing of the next series of bonds proceeds in an essentially periodic pattern because each ruptured bond has the same environment. When the remaining helical segment is short enough, failure affects bonds across its entire length, leading to two higher peaks. The native contacts are also rotated with respect to the pulling force at this stage, and this increases the force needed to rupture them. In the final stages (i.e., d > 28 Å), all the coils have been broken, and the series of small force peaks is due to breaking of local contacts between beads separated by 2 and 3 along the chain. These do not correspond to hydrogen bonds.

When a soft spring is used, the drop in force due to each event is smaller. If the threshold force for an event is lower than that for the previous event, the force may not drop below this threshold. This can cause several bond ruptures to accumulate into a single orchestrated event. The low velocity curve in the lower panel of Figure 3 has the same initial sequence of peaks as the top panel: Two small peaks are followed by several at the same higher force. However, those later peaks that are well below preceding peaks in the top panel are absent in the bottom panel. The strength of the contacts broken in these stages would be difficult to extract if a soft cantilever were used.

When the pulling velocity is comparable to the rapid motions produced by bond rupture, the cantilever motion can produce a substantial change in force during an unfolding event. This can also cause events to accumulate as shown in both panels. The increase in speed also produces a larger drag force from the surrounding solution (represented by the Langevin damping). This shifts the force curves to higher values.

The optimal temperature for folding of H16 has been established to be $T_{\rm min} = 0.3 \ \epsilon/k_B$.²⁶ The sequencing of thermal folding and unfolding events at $T_{\rm min}$ is shown in Figures 4 and 5. The former figure considers establishment of the contacts of the i,i+4 type, i.e., the hydrogenbonded contacts, whereas the latter is for the i,i+3 contacts. The time for establishing a given contact is denoted by t_c . These times are symmetrically arranged around the center of the helix and are shortest at the ends. We have also determined times for thermal unfolding, t_u , defined as times at which the contact is gone for the first time. Values of t_u in Figures 4 and 5 are averaged over 1500 different trajectories, which all start in the native state. We notice

108



Fig. 4. Sequencing of events as measured through forming or breaking bonds of the *i*,*i*+4 kind. The *x*-axis shows *i*, the monomer number along the chain. The moving cantilever is attached to *i* = 16. The *y*-axis shows *t_c*, *t_u*, and *d_u*. The first quantity is the mean time needed to establish the contact at T_{min} (based on 1000 trajectories). The second quantity is the first unfolding time at T_{min} for a bond under the conditions of no pulling force (based on 1500 trajectories). The third quantity, denoted by the circles, is the displacement where the contact is broken during mechanical unfolding at T = 0. Solid circles, slow pulling by the stiff spring; open circles, slow pulling by the soft spring. Values of *d_u* for the soft cantilever are divided by 2. Overall, the size of the symbol is a measure of the error bars, and all lines are guides to the eye.

that t_u is longer than t_c , but it is also arranged symmetrically around the center of the helix.

Figures 4 and 5 also show the displacements, d_u , at which each bond ruptures during mechanical unfolding for stiff (closed circles) and soft (open circles) cantilevers. These curves do not have the same symmetry as the thermal folding and unfolding curves. As noted above, the end bonds break first because they have fewer native contacts. Subsequent bonds have the same number of contacts and should break at the same force. However, the bonds near the pulling end (large i) tend to break first because of the presence of a small extra drag force. This is independent of the nature of the cantilever, except that the soft spring yields uniformly larger d_u at which a bond breaks.

Despite the lack of symmetry in the mechanical data, the contact formation times and contact breaking distances are clearly correlated. This is shown in Figures 6 and 7 for the stiff and soft springs, respectively. In each figure, contacts folding at later times tend to break at larger displacements.



Fig. 5. Same as in Figure 4, but for the bonds of the i,i+3 kind. Values of d_i , for the soft cantilever are divided by 2.

Two Helices in Tandem

We now consider two H16 helices connected in series by one extra peptide bond and stretched from one end. Figure 8 shows a snapshot of a partially unfolded tandem conformation. It indicates that the two helices unfold simultaneously with some phase shift between them. This is also seen in the F versus d curves shown in Figure 9, where the stick-slip patterns essentially double each feature seen in Figure 3. This behavior is quite distinct from what happens when stretching titin, where the domains unfold one at a time.²³ The basic reason that the helices unfold simultaneously is that the force to break contacts rises smoothly during the unfolding process. The heights of the force peaks only drop in the very late stages of growth when the coils are all gone. In the case of titin, one of the early peaks is higher than subsequent peaks. Once this contact breaks in one of the repeat units, there is a series of weaker bonds that can continue to rupture within that unit. These contact failures keep the force from rising back to a level that would initiate failure of the strong bonds in other repeat units.

The simultaneous unwinding of the two helices is also seen in Figure 10, which is an analog of Figure 4 for the single helix (minus the data on thermal unfolding). The distance for contact rupture (for i,i+4 contacts) through stretching shows two skewed peaks, each centered in the vicinity of the centers of the individual helices. In contrast, the average times for unfolding at $T = 0.3 \epsilon/k_B$ are peaked



Fig. 6. Stretching distances at which a bond rupture takes place (from Figs. 4 and 5) plotted versus average time needed to establish contact on folding. This is the case of a stiff spring that is being pulled slowly.



Fig. 7. Same as in Figure 6 but for the soft pulling spring.

not at the centers of the helices but at the very center of the whole system, i.e., around the peptide bond that connects the helices. Thus, the simple correlation between d_u and t_c that was seen in Figures 6 and 7 is lost. Instead one finds a two-legged correlation that is shown in Figure 11. Note



Fig. 8. Conformation of two α -helices connected in series after moving the cantilever by the distance indicated.



Fig. 9. Force versus displacement for H16-2: two α -helices H16 connected in series and pulled slowly.

also that all of the contacts (all are short ranged and are grouped into three types: i,i+2, i,i+3, and i,i+4) break throughout the full range of the displacement of the cantilever. Some bonds of a given kind break early; some break late. We shall see in the companion article²³ that failure of long range bonds shows a definite correlation with the displacement.

β-hairpin

The stretching of the β -hairpin B16, shown in Figure 12, consists of a gradual removal of the "rungs" of the "ladder" that form the hairpin, starting from the free ends. Physically, these rungs represent hydrogen bonds, and they correspond to contacts 1-16, 2-15, 3-14, ..., 7-10. There are



Fig. 10. Sequencing of events in H16-2 as measured through the bonds of the i,i+4 kind. 1000 trajectories were used in the studies of folding.

other contact forces in our Go model, and they provide further stabilization of the structure. These other bonds bind bead 1 with bead 15, bead 2 with 14 and 16, etc. As for the α -helix, bonds between native contacts are roughly parallel to the external force during unfolding. However, the distribution of tensile stress among the bonds is very different. The last surviving rung of the β -hairpin carries almost all of the strain, whereas all bonds are stressed in the α -helix. Thus, the order of bond breaking in the β -hairpin is determined by geometry rather than the relative strength of the bonds.

Plots of F versus d during unfolding at low v_p are shown in Figure 13. All show regular stick-slip features. In this respect, our results are very similar to those obtained by Bryant et al.¹⁹ with full atom simulations. Thus, our simplified model reproduces the features present in the more realistic calculation. Furthermore, because our model incorporates the native conformation but not the hydrophobic or polar properties of the amino acids, we suggest that the latter are not explicitly crucial in the mechanical unfolding of the hairpin. The stiff and soft springs produce the same sequence of stick-slip peaks, but the slope of the ramps and depth of the drops are smaller for the soft spring. After the first peak, peaks come in pairs, where the second peak has a lower height. When a very soft spring is used, these pairs merge into single large events as described above.



Fig. 11. Stretching distances at which a bond rupture in H16-2 takes place versus average time needed to establish contact on folding. The cantilever is stiff and pulled slowly. The contacts corresponding to the i,i+2 type are shown by the asterisks. The remaining symbols differentiate between the contacts present in the first helix (circles) and those present in the second helix (squares), the one that is closer to the cantilever. Open circles and squares, contacts of the i,i+3 type; filled circles and squares, the i,i+4 type.

The folding properties of B16 are illustrated in Figure 14. This system has been studied in detail in Reference 26, where the native basin has been accurately determined through a "shape distortion technique,"³⁹ which produces $T_{\rm min}$ of order 0.07 ϵ/k_B . If the folding criterion is based on just establishing the native contacts, then, in the case of B16, there is a very broad dependence of the folding time on temperature and the kinetics of folding at 0.07 ϵ/k_B is almost the same as at, say, 0.3 ϵ/k_B . Nevertheless we study the system at the previously determined T_{min} . Note that even with the contact-based criterion for folding, the folding time for B16 is still considerably longer than for H16.

Figure 14 shows that the sequencing of folding events in B16 is exactly opposite to the succession of contact breakage upon stretching: B16 starts folding from the turn (the result that has been found both experimentally²⁰ and theoretically, ^{26,40} whereas both mechanical and thermal unfolding start at the free ends. Thus, in contrast to the α -helix, the mechanical unfolding of the β -hairpin is the inverse of the folding process.

Figure 15 shows d_u as a function of the time needed to establish the contact during folding. Here, in addition to the "rung" contacts, the remaining contacts are also shown. Because contacts rupture at a fixed force, the soft spring data are shifted to larger displacements than the the stiff spring data. However, both sets of data show a clear



Fig. 12. Snapshot pictures of stretching of the β -hairpin B16 for *d* equal to 5 and 20 Å. All bonds between native contacts are roughly aligned with the pulling force, but most of the stress is concentrated in the bond nearest the pulling springs.



Fig. 13. The force versus displacement curves for the β -hairpin B16 obtained at slow pulling velocities for the indicated cantilever stiffnesses.



Fig. 14. Similar to Figure 4 but for the hydrogen rung-like bonds in B16. The bonds are identified by the index *i* that they connect to. The thermal data are based on 1000 trajectories and are collected at $T_{min} = 0.07 \epsilon/k_B$. The flat character of the data corresponding to thermal unfolding is expected to turn into a steeper dependence at higher temperatures.

anticorrelation between thermal folding and mechanical unfolding that is in a sharp contrast to the results for the α -helix.

Irreversibility Length

We now consider pulling of a protein at a constant slow rate and then releasing it. We ask what is the time required to fold back to the native state at T = 0. There must be a limit to the extension beyond which the protein misfolds on release. Figure 16 shows that this limit indeed exists. The dependence on cantilever stiffness is minimized by plotting the refolding times against the end-toend distance L of the protein rather than the cantilever displacement. For both stiffnesses the refolding times are found to be non-monotonic functions of L. We interpret this as being due to inertial effects. The more stretched the protein is with a given set of contacts, the more potential energy is available. When the protein is released, the energy is converted into kinetic energy that speeds the contraction of the protein and aids it in getting over subsequent energy barriers.

We identify the irreversibility length L_{ir} with the maximum value of L where refolding occurs. For H16, L_{ir} is about 37 Å, or 1.6 times the native state end-to-end distance of 22.62 Å. The change in length is 14.4 Å which is very close to the displacement of the stiff cantilever at the onset of irreversibility $d_{ir} = 14.9$ Å. The displacement of



Fig. 15. Stretching distances at which a bond rupture in B16 takes place versus average time needed to establish contact on folding. The cantilever is pulled slowly. Solid circles, the "rung" contacts; open circles, the remaining contacts. There are degeneracies related to these other contacts. For instance, 2-14 forms and breaks at essentially the same moment (statistically) as 3-15. Examples of other such pairs are 3-13 with 4-14 and 6-12 with 5-11.

the soft cantilever, $d_{ir} = 37.4$ Å, is larger because the cantilever stretches more in order to apply enough force to reach L_{ir} . Examining Figure 3, we see that both values of d_{ir} correspond to the displacement after the sixth peak in the respective force curve. Thus, the same set of broken bonds is required to produce irreversibility for either cantilever stiffness.

For B16, the native L is only 5 Å, and the stretching factor to L_{ir} is substantially larger, ~ 11.6 . The values of d_{ir} , for stiff and soft cantilevers are $d_{ir} = 52.9$ and 65.9 Å, respectively. From Figure 13 we see that in both cases the irreversibility point is just past the last peak in the force curve. Because the protein is fully stretched at this point, any native contacts are enough to ensure refolding.

The misfolded conformations that are obtained on refolding beyond the threshold are shown in Figure 17. In the case of B16, the turn region freezes into the wrong configuration, which is almost straight. In the case of H16, the first turn coils with the wrong chirality.

CONCLUSIONS

We have studied the force-displacement curves for secondary structures of proteins for two models of cantilever stiffness and several pulling speeds. A series of stick-slip events is observed as contacts break. Stiff cantilevers pulled at low rates provide the most detailed information about the breaking of individual contacts. Multiple contact ruptures merge into single events when the stiffness is decreased or the speed is increased.



Fig. 16. Refolding times after stretching to the indicated end-to-end distance. The main figure is for B16 and the inset for H16. Solid lines, pulling by a stiff cantilever; dotted lines, pulling by a soft cantilever. The curves end at L_{μ} . To the right of the data points shown, the protein does not return to its native state. For B16, the corresponding threshold values of the tip displacement, d_{μ} are equal to 52.9 and 65.9 Å for the stiff and soft cantilevers are 14.9 and 37.4 Å, respectively.



Fig. 17. Conformations corresponding to the misfolded proteins after stretching just beyond the irreversibility threshold.

The simple expectation that mechanical unraveling should proceed in the inverse order from thermal folding is only confirmed in the case of the β -hairpin. In the case of the α -helix, unraveling and folding follow the same order. When multiple helices are connected in tandem, the correlation becomes even more complex. The two helices unravel simultaneously with each helix uncoiling from both of its ends. In contrast, folding occurs first at the outer ends of the pair of helices.

The differences in behavior of the two simple proteins considered here result from differences in connectivity and geometry. Stress is distributed among all bonds along the α -helix, and the order of breaking is determined by the relative bond strength. The relative bond strength is unimportant for the β -hairpin, because stress is concentrated on the terminal bond. Similar stress concentrations are likely at any bonds that hold the ends of a loop together.

Another factor that will be important in more complex proteins is the orientation of the bond relative to the applied force.²² The bonds in the simple proteins considered here were almost always aligned with the pulling force. In other proteins the orientation may vary with position and time. Such changes in orientation can lead to higher breaking forces like those seen in the final stages of unfolding of the α -helix. In general there is no reason to expect a simple correlation between thermal folding and mechanical unfolding of proteins. In the companion article,²³ we examine similar issues of mechanical-thermal correlations for a protein with a significant number of long-ranged contacts.

ACKNOWLEDGMENTS

The authors appreciate discussions with J.R. Banavar and T. Woolf, which helped in motivating this research.

REFERENCES

- Leckband D. Measuring the forces that control protein interactions. Annu Rev Biophys Biomol Struct 2000;29:1–26.
- Willemsen OH, Snel MME, Cambi A, Greve J, De Grooth BG, Figdor CG. Biomolecular interactions measured by atomic force microscopy. Biophys J 2000;79:3267–3281.
- Heinz WF, Hoh JH. Spatially resolved force spectroscopy of biological surfaces using the atomic force microscope. Trends Biotechnol 1999;17:143-150.
- Heinz WF, Antonik MD, Hoh JH. Reconstructing local interaction potentials from perturbations to the thermally driven motion of an atomic force microscope cantilever. J Phys Chem B 2000;104:622– 626.
- Evans E. Probing the relation between force-lifetime-and chemistry in single molecular bonds. Annu Rev Biophys Biomol Struct 2001;30:105–128.
- Kellermayer MSZ, Smith SB, Granzier HL, Bustamante C. Foldingunfolding in single titin molecules characterized with laser tweezers. Science 1997;276:1112–1116.
- Tskhovrebova L, Trinick K, Sleep JA, Simmons M. Elasticity and unfolding of single molecules of the giant muscle protein titin. Nature 1997;387:308–312.
- Israelachvili J, Wennerstrom H. Role of hydration and water structure in biological and colloidal interactions. Nature 1996;379: 219–215.
- Evans E, Ritchie K, Merkel R. Sensitive force technique to probe molecular adhesion and structural linkages at biological interfaces. Biophys J 1995;68:2580-2587.
- Reinisch L, Kaiser RD, Krim J. Measurement of protein hydration shells using a quartz microbalance. Phys Rev Lett 1989;63:1743– 1746.
- Keller CA, Glasmaestar K, Zhdanov VP, Kasemo B. Formation of supported membranes from vesicles. Phys Rev Lett 2000;84:5443– 5446.
- Williams MC, Rouzina I, Wenner JR, Gorelick RJ, Musier-Forsyth K, Bloomfield VA. Mechanism for nucleic acid chaperone activity of HIV-1 nucleocapsid protein revealed by single molecule stretching. Proc Natl Acad Sci USA 2001;98:6121–6126.
- Bockelmann U, Essevaz-Roulet B, Heslot F. Molecular stick-slip motion revealed by opening DNA with piconewton forces. Phys Rev Lett 1997;79:4489-4492.
- 14. Yang G, Cecconi C, Baase WA, Vetter IR, Breyer WA, Haack JA,

Matthews BW, Dahlquist FW, Bustamante C. Solid-state synthesis and mechanical unfolding of polymers of T4 lysozyme. Proc Natl Acad Sci USA 2000;97:139–144.

- Abe H, Go N. Noninteracting local-structure model of folding and unfolding transition in globular proteins. II. Application to twodimensional lattice proteins. Biopolymers 1981;20:1013–1031.
- Takada S. Go-ing for the prediction of protein folding mechanism. Proc Natl Acad Sci USA 1999;96:11698–11700.
- 17. Pearlman DA, Case DA, Caldwell JW, Ross WS, Cheatham TC, Debolt SE, Ferguson DM, Seibel GL, Kollman PA. AMBER, a package of computer programs for applying molecular mechanics, normal mode analysis, molecular dynamics and free energy calculations to simulate the structural and energetic properties of Molecules. Comp Phys Comm 1995;91:1–41.
- Brooks BR, Bruccoleri RE, Olafson BD, States DJ, Swaminathan S, Karplus M. CHARMM: a program for macromolecular energy minimization and dynamics calculations. J Comput Chem 1983;4: 187–217.
- Bryant Z, Pande VS, Rokhsar DS. Mechanical unfolding of a β-hairpin using molecular dynamics. Biophys J 2000;78:584–589.
- Munoz V, Thompson PA, Hofrichter J, Eaton WA. Folding dynamics and mechanism of β-hairpin formation. Nature 1997;390:196– 199.
- Munoz V, Henry ER, Hofrichter J, Eaton WA. A statistical mechanical model for β-hairpin kinetics. Proc Natl Acad Sci USA 1998;95:5872–5879.
- Socci ND, Onuchic JN, Wolynes PG. Stretching lattice models of protein folding. Proc Natl Acad Sci USA 1999;96:2031–2035.
- Cieplak M, Hoang TX, Robbins MO. Folding and stretching in a Go-like model of titin. Proteins Struct Funct Genet 2002;49:114– 124.
- Socci ND, Onuchic JN. Folding kinetics of protein-like heteropolymers. J Chem Phys 1994;101:1519–1528.
- Cieplak M, Henkel M, Karbowski J, Banavar JR. Master equation approach to protein folding and kinetic traps. Phys Rev Lett 1998;80:3654-3657.
- Hoang TX, Cieplak M. Molecular dynamics of folding of secondary structures in Go-like models of proteins. J Chem Phys 2000;112: 6851–6862.
- Hoang TX, Cieplak M. Sequencing of folding events in Go-like proteins. J Chem Phys 2001;113:8319-8328.
- Cieplak M, Hoang TX. Kinetics non-optimality and vibrational stability of proteins. Proteins Struct Funct Genet 2001;44:20-25.
- Unger R, Moult J. Local interactions dominate folding in a simple protein model. J Mol Biol 1996;259:988–994.
- Plaxco KW, Simons KT, Baker D. Contact order, transition state placement and the refolding rates of single domain proteins. J Mol Biol 1998;277:985–994.
- Plaxco KW, Simons KT, Ruczinski I, Baker D. Topology, stability, sequence, and length: defining the determinants of two-state protein folding kinetics. Biochemistry 2000;39:11177-11183.
- Cieplak M, Hoang TX. Scaling of folding properties in Go models of proteins. J Biol Phys 2000;26:273–294.
- Clementi C, Maritan A, Banavar JR. Folding, design and determination of interaction potentials using off-lattice dynamics of model heteropolymers. Phys Rev Lett 1998;81:3287–3290.
- Clementi C, Vendruscolo M, Maritan A, Domany E. Folding Lennard-Jones proteins by a contact potential. Proteins Struct Funct Genet 1999;37:544-553.
- Veitshans T, Klimov D, Thirumalai D. Protein folding kinetics: time scales, pathways and energy landscapes in terms of sequencedependent properties. Folding Des 1997;2:1–22.
- Klimov DK, Thirumalai, D. Viscosity dependence of the folding rates of proteins. Phys Rev Lett 1997;79:317–320.
- 37. Gear WC. Numerical initial value problems in ordinary differential equations. New York: Prentice-Hall, Inc.; 1971.
- Lu H, Schulten M. The key event in force-induced unfolding of titin's immunoglobulin domains. Biophys J 2000;79:51–65.
- Li MS, Cieplak M. Delineation of the native basin in continuum models of proteins. J Phys A 1999;32:5577–5584.
- Klimov DK, Thirumalai D. Mechanisms and kinetics of β-hairpin formation. Proc Natl Acad Sci USA 2000;97:2544-2549.