

# Tuning physical properties of NiFe<sub>2</sub>O<sub>4</sub> and NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> nanoparticles by thermal treatment

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To read more: Bajorek, A., et al. Tuning Physical Properties of NiFe<sub>2</sub>O<sub>4</sub> and NiFe<sub>2</sub>O<sub>4</sub>@SiO<sub>2</sub> Nanoferrites by Thermal Treatment. Metall Mater Trans A 53, 1208–1230 (2022). <https://doi.org/10.1007/s11661-021-06567-0>

## Summary

The comparison between NiFe<sub>2</sub>O<sub>4</sub> (co-precipitation) and NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> (co-precipitation and microemulsion) ferrite nanoparticles in their as-received and annealed form is presented. The structural characterization revealed the gradual crystallization of as-received samples induced by thermal treatment. The existence of cubic inverse spinel ferrite structure with tetrahedral and octahedral iron occupancy is confirmed in all samples by the comprehensive study. In the case of nanoparticles embedded into the silica matrix, the crystallization of initially amorphous silica is revealed in structural and microstructural characterization. The separation of the rhombohedral hematite  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> phase in the NiFe<sub>2</sub>O<sub>4</sub> ferrite evidenced during the annealing process is demonstrated in structural and magnetic studies. The room temperature superparamagnetic state (SPM) is modified in the NiFe<sub>2</sub>O<sub>4</sub> sample by annealing as an effect of ferrite crystallization and grain growth as well as hematite separation. For as-received NiFe<sub>2</sub>O<sub>4</sub>, with temperature decrease, the blocking process preceded by the freezing process is observed. The silica shell is recognized as the sustaining cover for the SPM state. The electronic structure studies confirmed the complex nature of the Fe-based states.

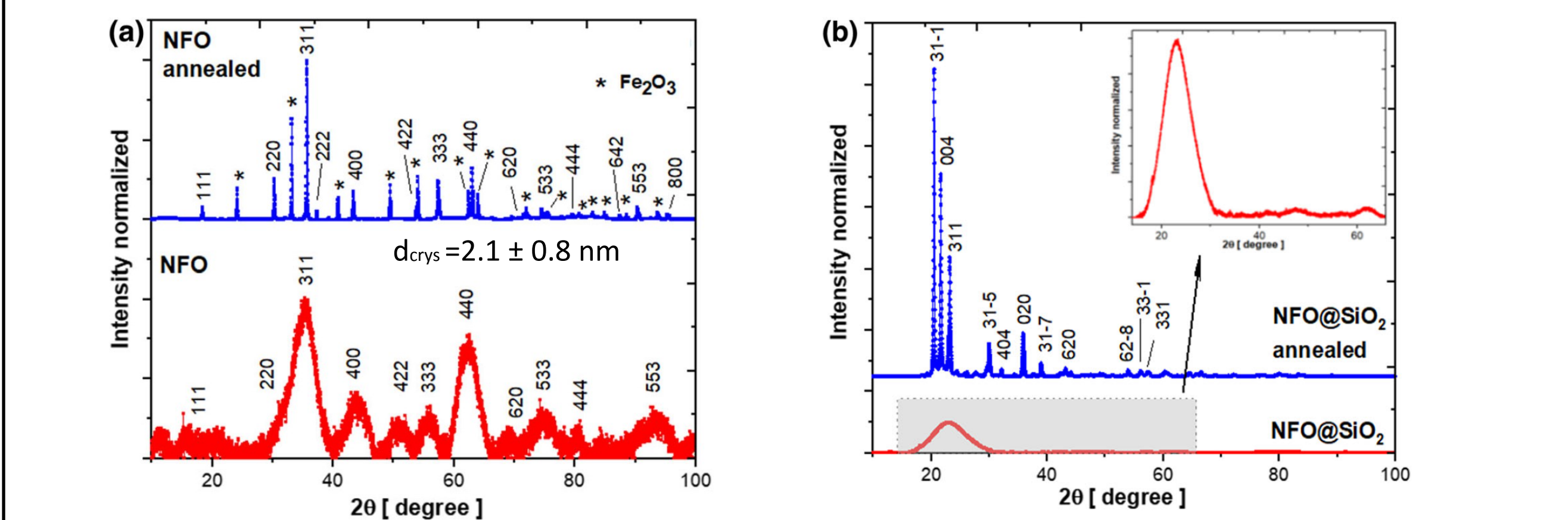


Fig. 1. XRD patterns for the investigated samples.

## NiFe<sub>2</sub>O<sub>4</sub>

- cubic inverse spinel structure (*Fd3m*),  $d \sim 2$  nm
- the diffraction peaks are broadened, indicating the ultrafine crystal structure

## NiFe<sub>2</sub>O<sub>4</sub> annealed

- 52% NiFe<sub>2</sub>O<sub>4</sub> → cubic inverse spinel structure (*Fd3m*)  $d \sim 58$  nm
- 48% rhombohedral hematite  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> (*R-3c*)  $d \sim 91$  nm

## NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub>

- spectrum dominated by the broad peak → tetragonal SiO<sub>2</sub> structure (amorphous silica matrix)
- trace of crystalline NiFe<sub>2</sub>O<sub>4</sub> nanoparticles covered by SiO<sub>2</sub>

## NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> annealed

- SiO<sub>2</sub> crystallized in the monoclinic *Aa* structure ( $d \sim 77$  nm)
- no evidence of NiFe<sub>2</sub>O<sub>4</sub>

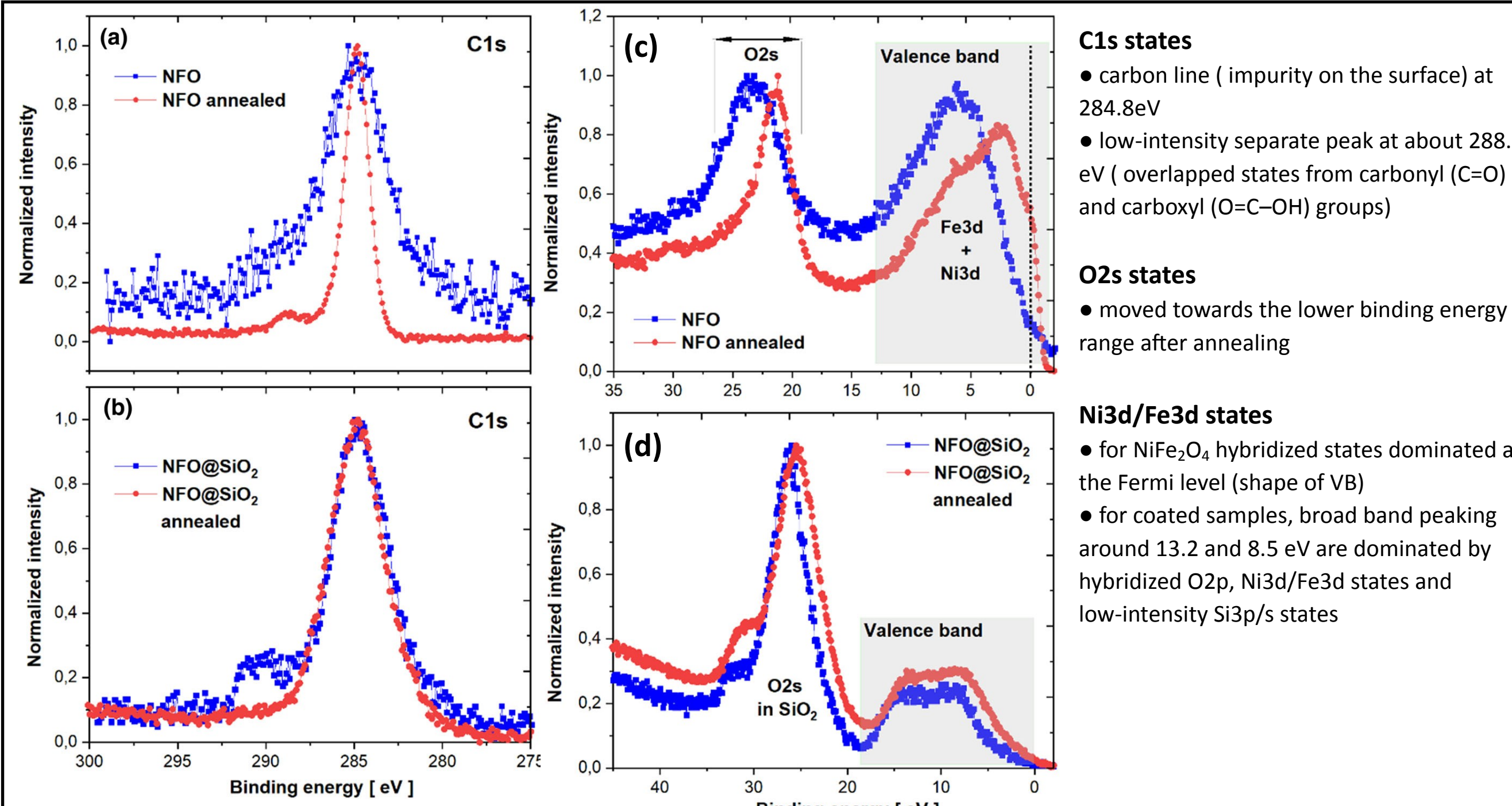


Fig. 6. Comparison between C1s states (a) - (b), and between valence bands spectra (c) - (d).

## Temperature dependences of the magnetization (VSM measurements)

- zero field cooled - field cooled regime
- $H_{ext} = 100$  Oe, 1000 Oe

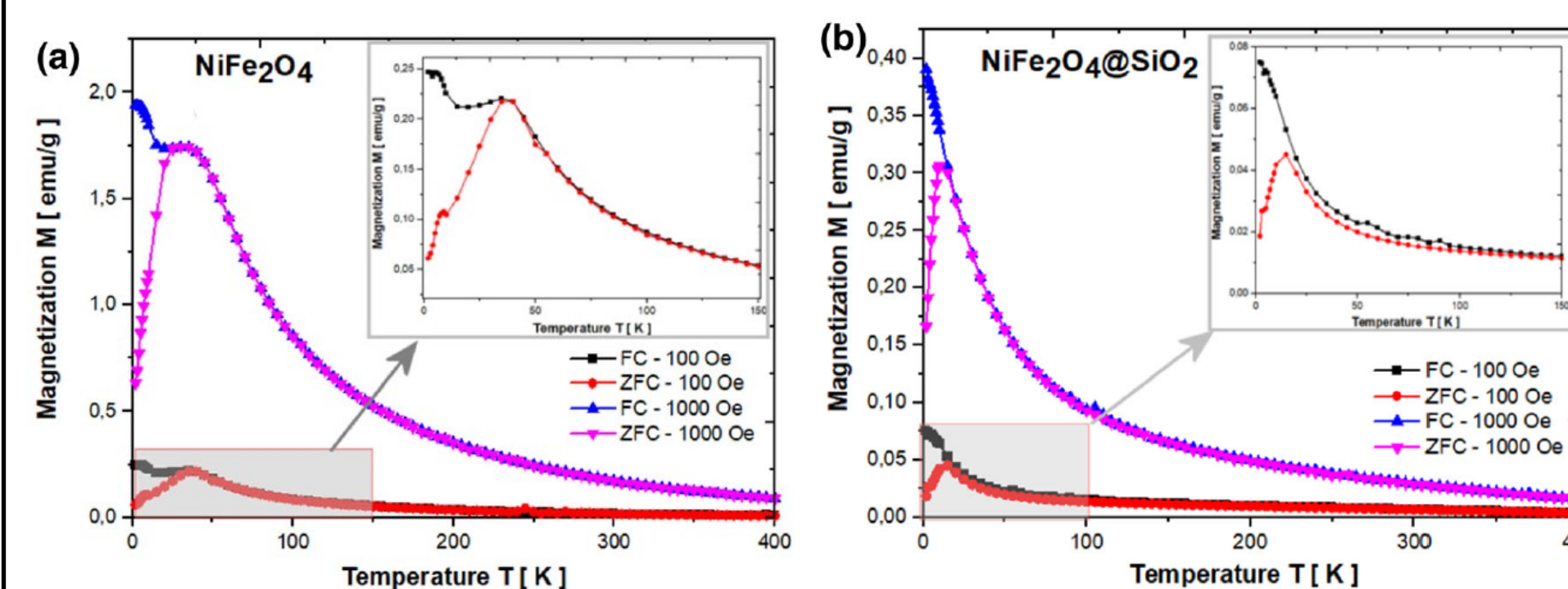


Fig. 8. ZFC-FC curves for the as-received samples.

## NiFe<sub>2</sub>O<sub>4</sub>

- two maxima:  $T_{max1} \sim 32$  K, and  $T_{max2} \sim 7$  K → the freezing process is observed before the blocking process
- the reversibility of ZFC-FC curves from  $T_{max1}$  to 400 K → superparamagnetic behavior

## NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub>

- the ZFC maximum at  $T_{max} = 14.8$  K (100 Oe) and  $\approx 11.9$  K (1000 Oe) → blocking process
- NiFe<sub>2</sub>O<sub>4</sub> particles are well dispersed in the silica matrix → the interaction between them is negligible

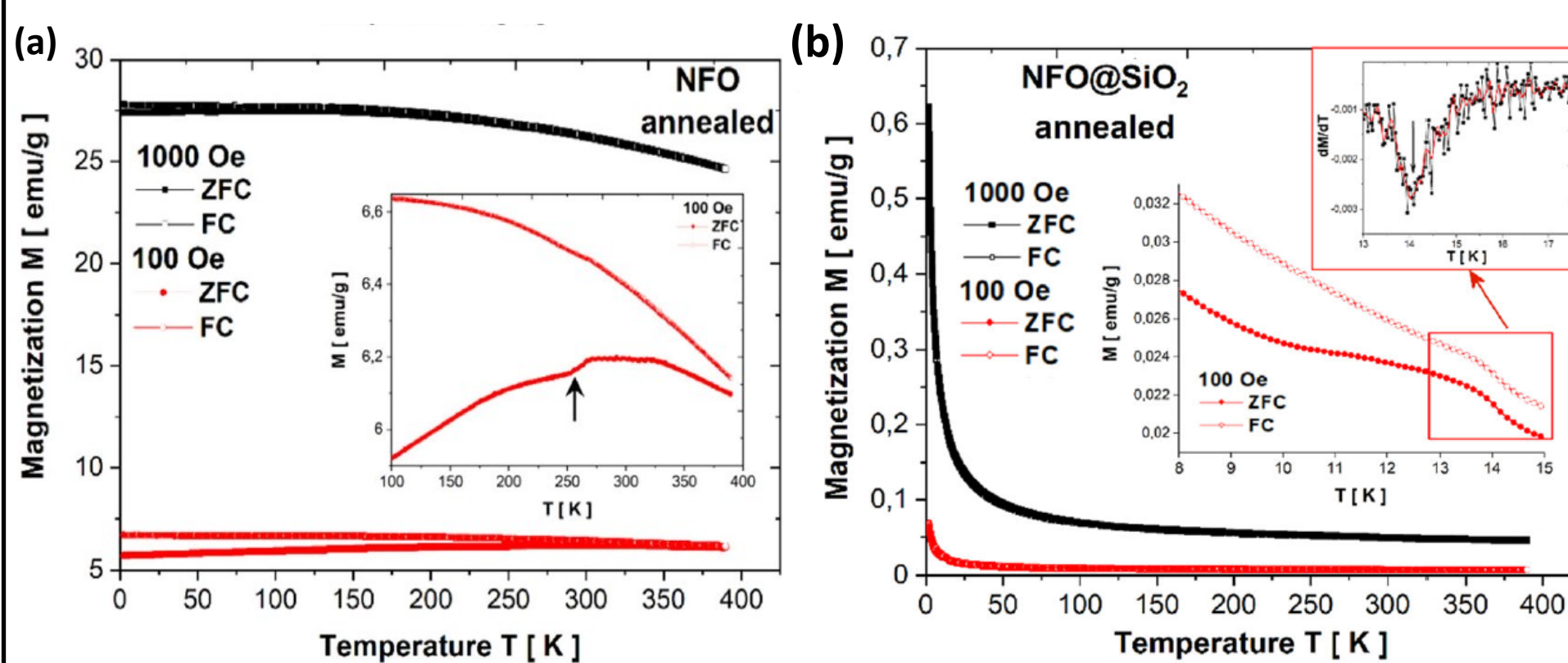


Fig. 9. ZFC-FC curves for the annealed samples.

## NiFe<sub>2</sub>O<sub>4</sub> annealed

- blocking or freezing processes are excluded → system behaves as a bulk material
- a kink in the ZFC curve between 250 K and 260 K denotes the Morin transition

## NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> annealed

- ZFC and FC curves show the magnetization increases with decreasing the temperature as in the paramagnetic system → the decomposition of ferrite nanoparticles to paramagnetic ions due to the annealing process
- a broad maximum  $\sim 14$  K → residues of ferrite nanoparticles

## Magnetization curves (VSM measurements)

- field range  $\pm 90$  kOe
- $T = 2$  K, 100 K, 300 K

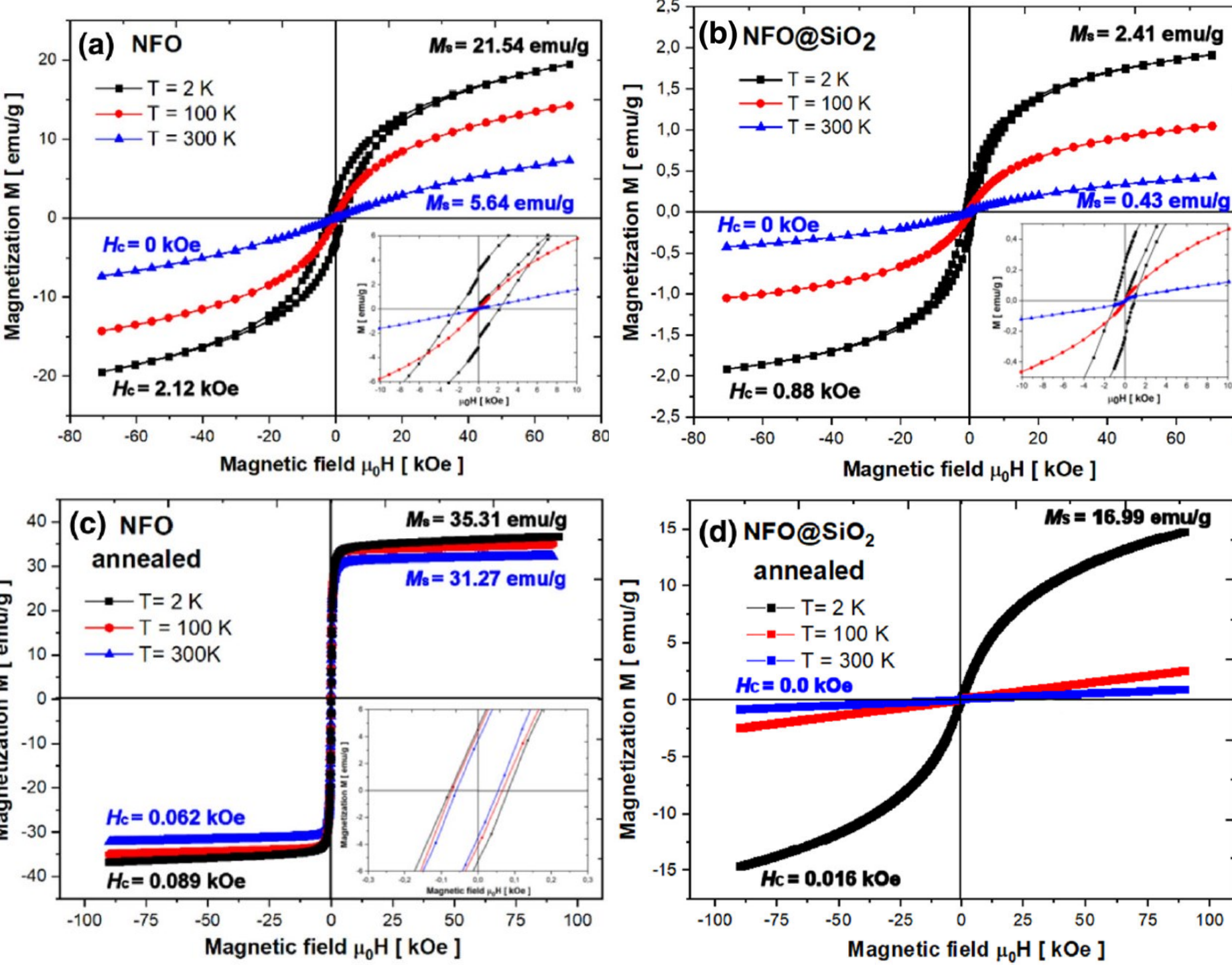


Fig. 10.  $M(H)$  curves for the investigated samples.

## Anisotropy

$$M(H) = M_s \left( 1 - \frac{a}{\sqrt{H}} - \frac{b}{H^2} \right) + \chi_p H$$

$$b \approx \frac{8}{105} \left( \frac{K_1}{\mu_0 M_s} \right)^2$$

$K_1$  is reduced with the increase of particle size

$K_1$ [ $10^5$ erg/cm <sup>3</sup> ]	$T = 2$ K	$T = 100$ K	$T = 300$ K
NiFe <sub>2</sub> O <sub>4</sub>	3.51	2.93	1.32
NiFe <sub>2</sub> O <sub>4</sub> coated with SiO <sub>2</sub>	0.36	0.25	0.09
NiFe <sub>2</sub> O <sub>4</sub> annealed	0.97	1.95	1.17
NiFe <sub>2</sub> O <sub>4</sub> coated with SiO <sub>2</sub> annealed	3.79	-	-

crystallization of silica shell and subsequent modification of NiFe<sub>2</sub>O<sub>4</sub> particle coating

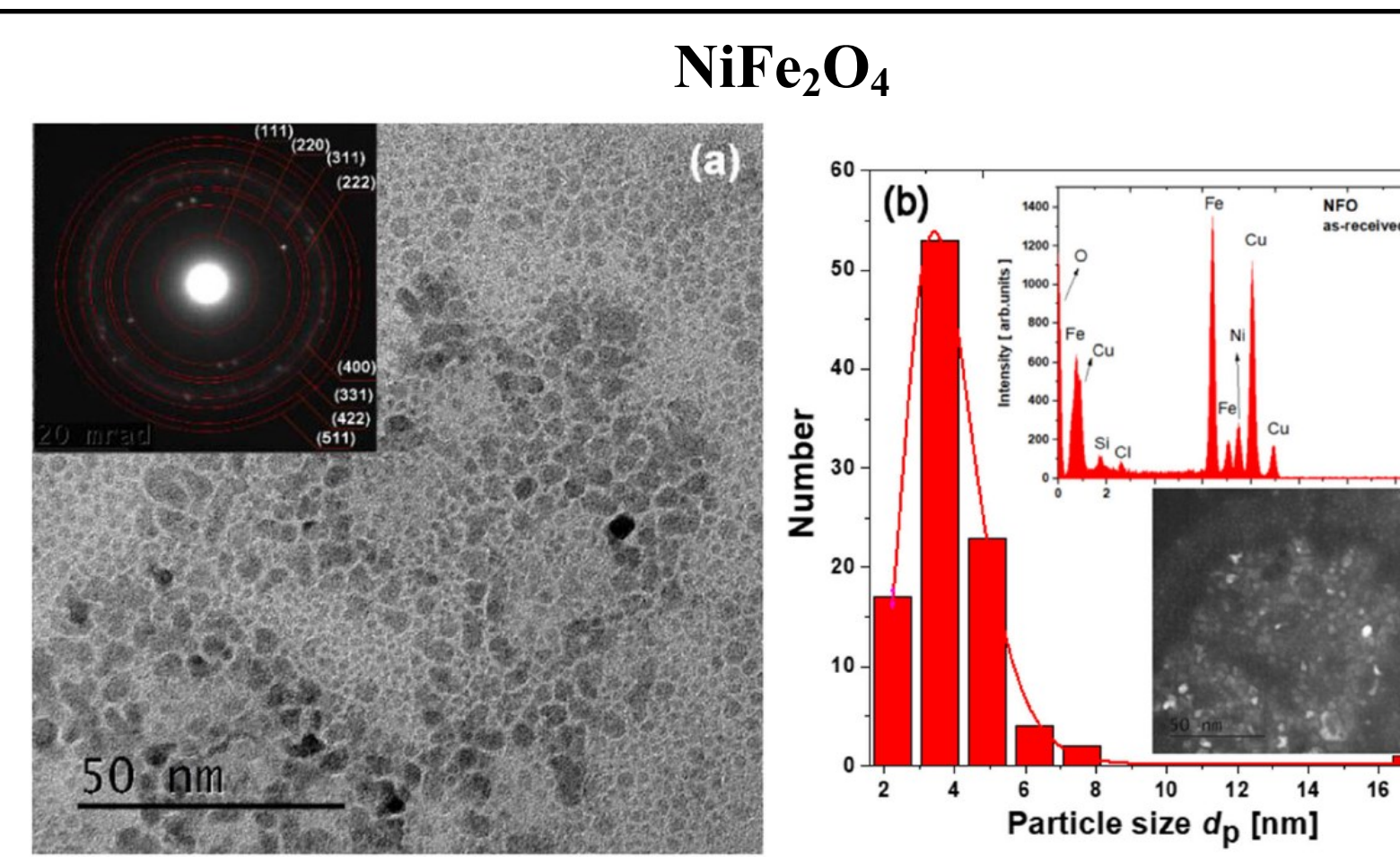


Fig. 2. (a) TEM micrograph and the SAED pattern for NiFe<sub>2</sub>O<sub>4</sub> → randomly oriented polycrystalline grains with inverse spinel symmetry. (b) Crystallite size distribution, and EDS analysis →  $d = (3.7 \pm 0.3)$  nm. The EDS spectrum confirmed the presence of the NiFe<sub>2</sub>O<sub>4</sub> phase.

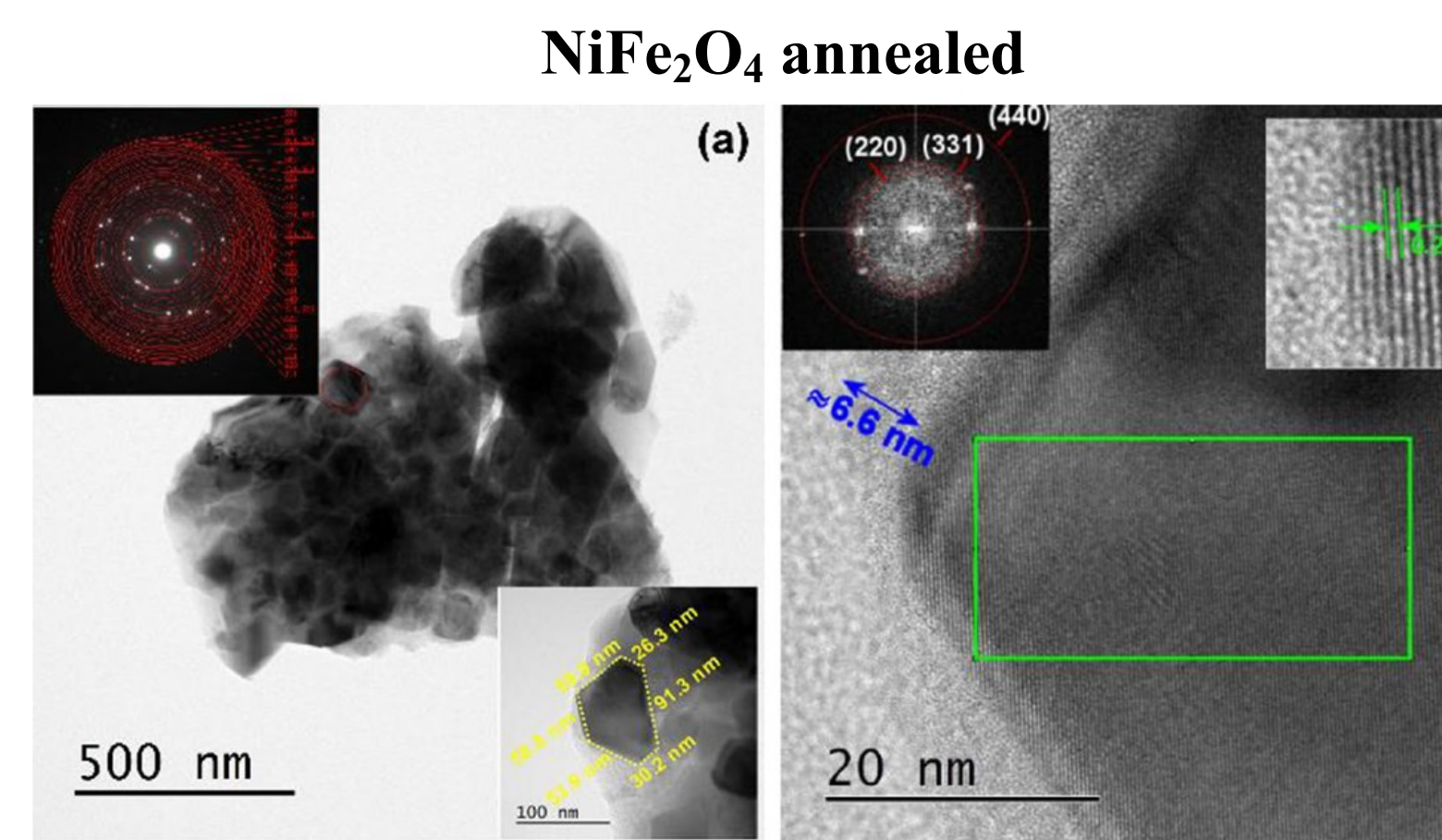


Fig. 4. (a) TEM micrograph and the SAED pattern for the annealed NiFe<sub>2</sub>O<sub>4</sub> nanoparticles → well-crystallized nanoparticles,  $d \sim 72$  nm, polygon shape. (b) The HR-TEM image of one crystallite → interplanar spacing of 0.26 nm between adjacent planes, corresponding to (311) NiFe<sub>2</sub>O<sub>4</sub> lattice plane.

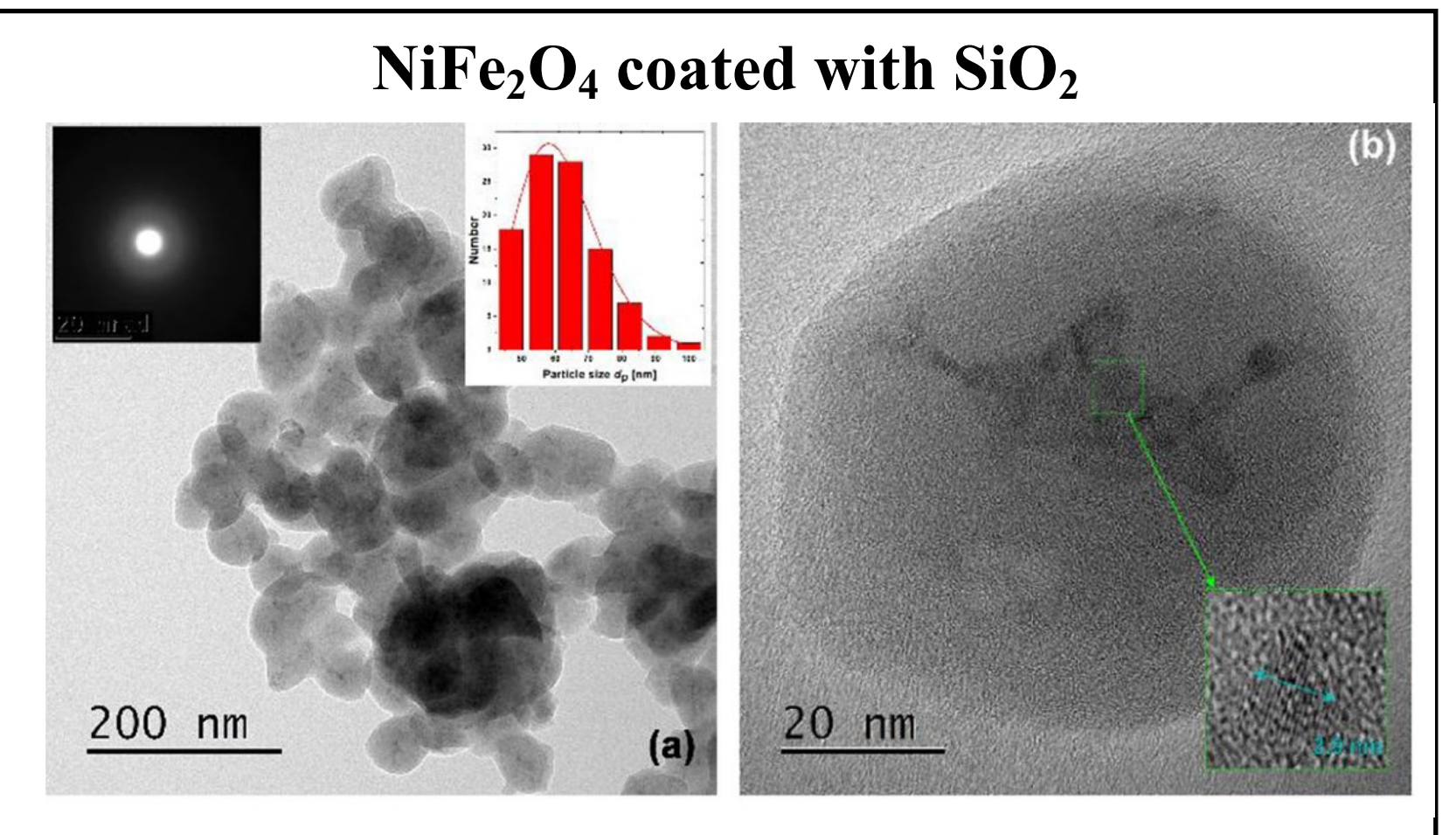


Fig. 3. (a) TEM micrograph and the SAED pattern for NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> → particles are almost spherical,  $d \sim 60$  nm. (b) The HR-TEM image → irregular distribution of NiFe<sub>2</sub>O<sub>4</sub> inside SiO<sub>2</sub>. The EDS spectrum confirmed the presence of the NiFe<sub>2</sub>O<sub>4</sub> phase.

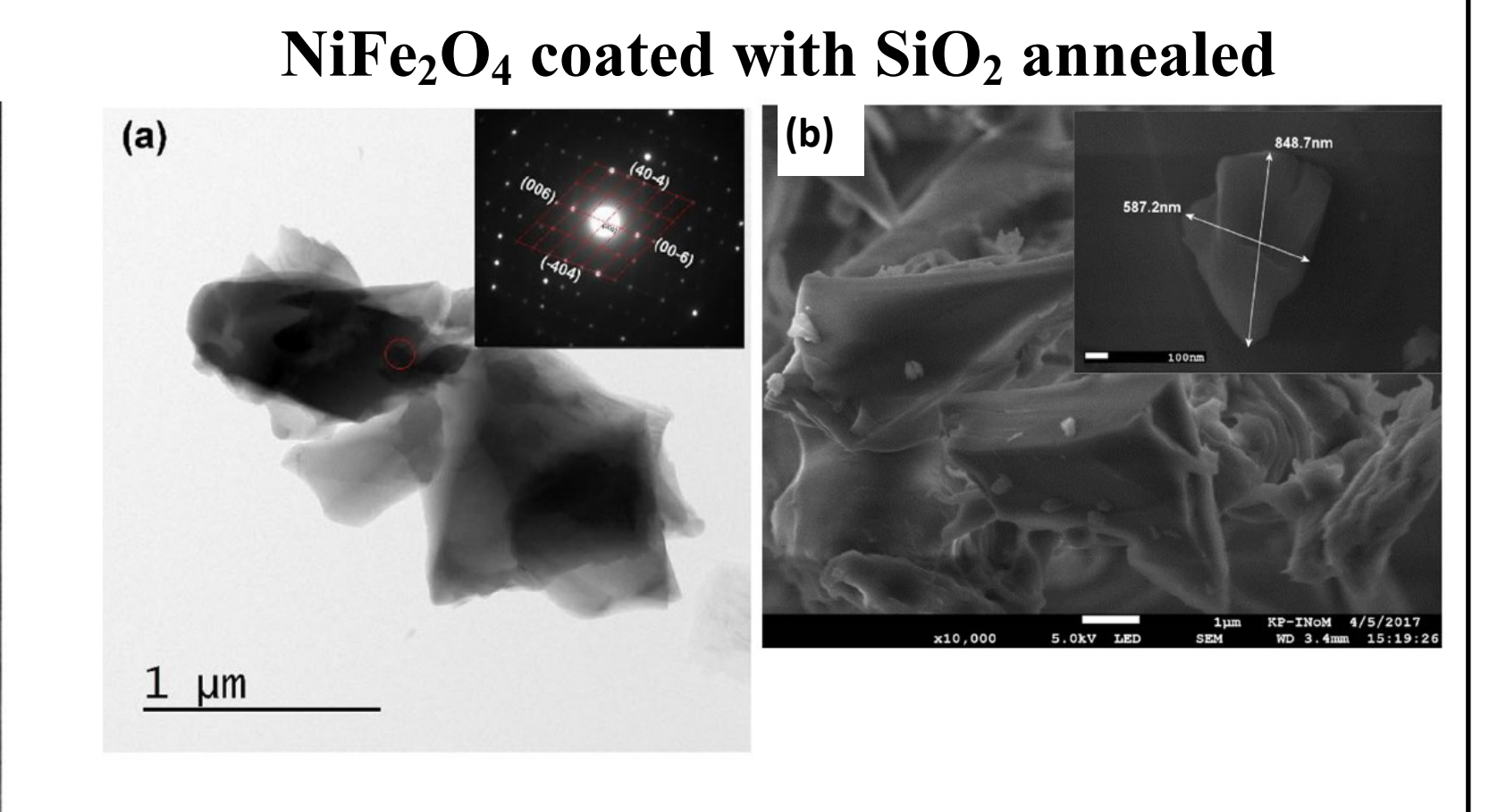


Fig. 5. (a) TEM micrograph and the SAED pattern for the annealed NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> → spherically shaped particles disappeared; thin irregular flakes (crystallized SiO<sub>2</sub>) have emerged. (b) The SEM image → large irregular grains, size  $\sim 600$  nm.

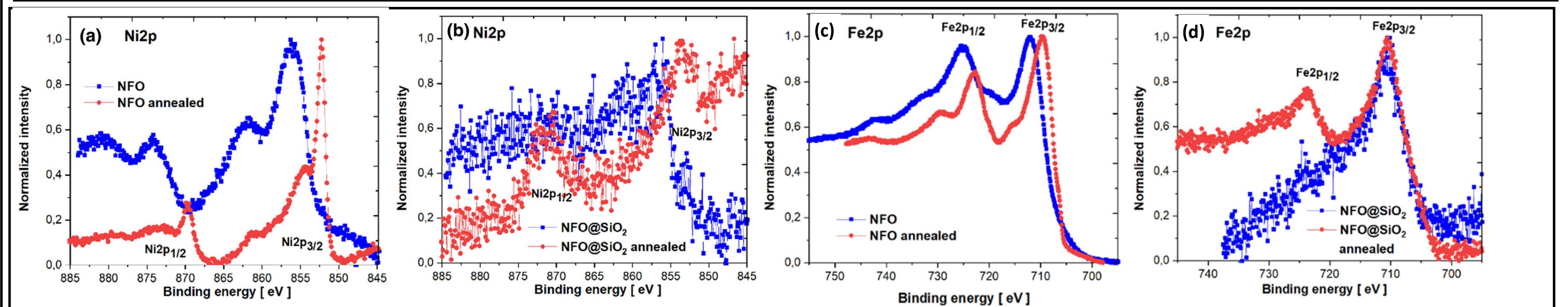


Fig. 7. Comparison between Ni2p Fe2p lines.

**NiFe<sub>2</sub>O<sub>4</sub>**  
two binding energy peaks: 854.7 eV (Ni2p<sub>3/2</sub>), 871.8 eV (Ni2p<sub>1/2</sub>)  
broad satellite peaks:  $\sim 862$  eV and 879 eV

**NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub>**  
Ni line barely visible → position similar to NiFe<sub>2</sub>O<sub>4</sub>

**NiFe<sub>2</sub>O<sub>4</sub>**  
Fe2p<sub>3/2</sub> and Fe2p<sub>1/2</sub> lines + satellites at 719.2 and 733.5 eV confirming the existence of Fe<sup>3+</sup> ions

**NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub>**  
weak Fe2p line

**NiFe<sub>2</sub>O<sub>4</sub> annealed**  
Ni2p<sub>3/2</sub> → metallic Ni line (852.4 eV) 16.2%, spinel ferrite line (853.7 eV) 40.72%, spinel ferrite line (854.3 eV) 31.2%,

**NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> annealed**  
Ni2p<sub>3/2</sub> → metallic Ni line (852.2 eV) 23.44%, spinel ferrite line (853.7 eV) 40.72%, 6eV satellites structure (855.28 and 856.36 eV)

**NiFe<sub>2</sub>O<sub>4</sub> annealed**  
lines are shifted towards a lower binding energy iron species exist in more than one chemical state (metallic Fe, and Fe<sub>2</sub>O<sub>3</sub>)

**NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> annealed**  
typical inverse spinel ferrite structure with the ratio of Fe<sup>2+</sup>(A)/Fe<sup>3+</sup>(B) = 0.8 and about 21.9% contribution of metallic iron

## Mössbauer spectra

- $T = 77$  K, 300 K

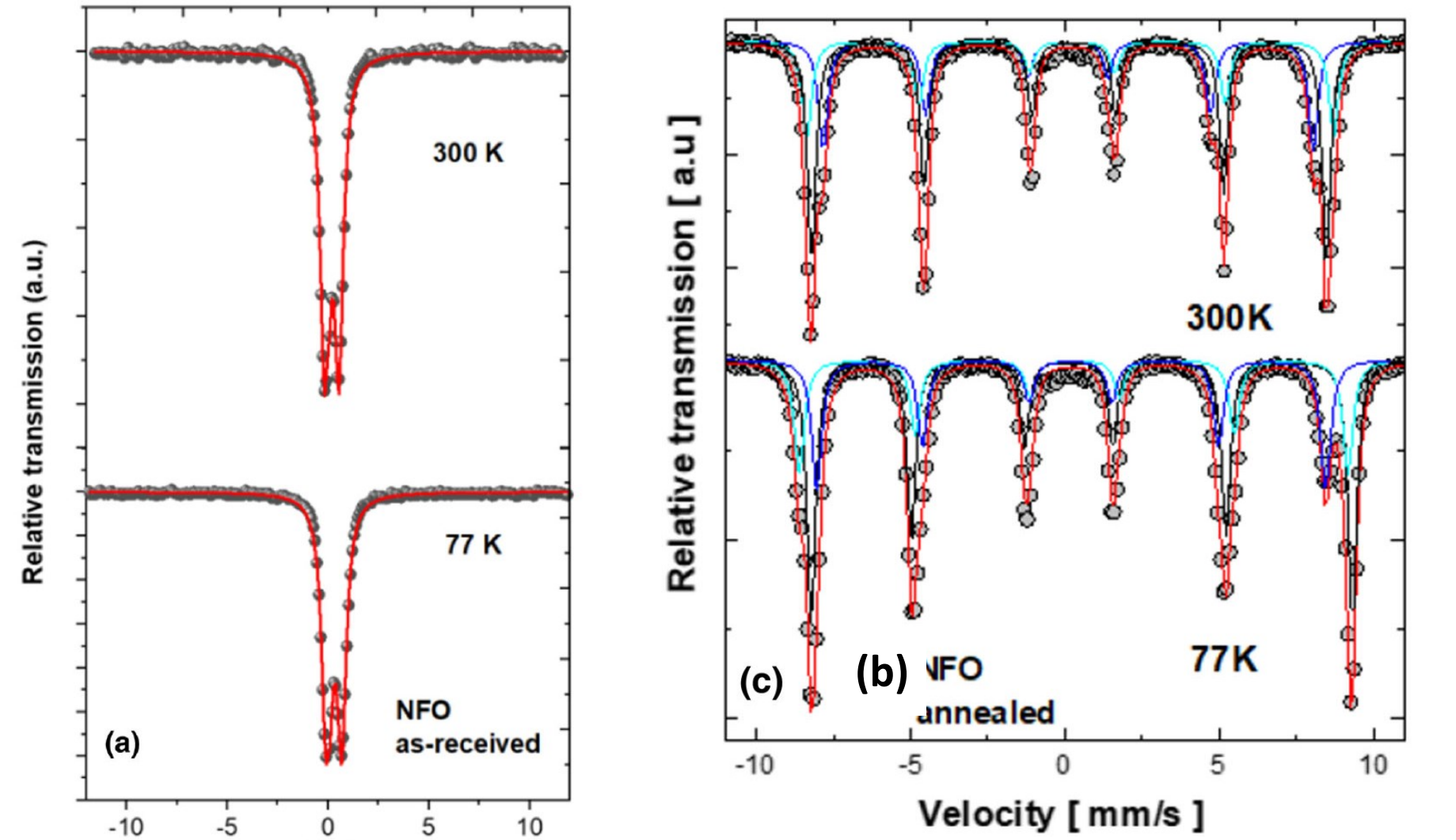


Fig. 11. Mössbauer spectra for NiFe<sub>2</sub>O<sub>4</sub> (a) and NiFe<sub>2</sub>O<sub>4</sub> annealed (b). The results for the samples coated with silicon are similar to (a).

	IS (mm/s)	QS/2ε (mm/s)	$B_{hyp}$ (T)	FWHM/2 (mm/s)	A (%)	Phase	
<b>NiFe<sub>2</sub>O<sub>4</sub></b>							
$T = 300$ K	0.32	0.63	-	0.26	100	NiFe <sub>2</sub> O <sub>4</sub>	
$T = 77$ K	0.44	0.67	-	0.27	100	NiFe <sub>2</sub> O <sub>4</sub>	
<b>NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub></b>							
$T = 300$ K	0.33	0.65	-	0.25	100	NiFe <sub>2</sub> O <sub>4</sub>	
$T = 77$ K	0.43	0.63	-	0.26	100	NiFe <sub>2</sub> O <sub>4</sub>	
<b>NiFe<sub>2</sub>O<sub>4</sub> annealed</b>							
$T = 300$ K	Site A	0.26	0	49.0	0.18	27	NiFe <sub>2</sub> O <sub>4</sub>
	Site B	0.36	-0.05	52.7	0.18	24	NiFe <sub>2</sub> O <sub>4</sub>
$T = 77$ K	Site A	0.36	0.02	50.9	0.19	24	NiFe <sub>2</sub> O <sub>4</sub>
	Site B	0.46	-0.04	54.9	0.19	28	NiFe <sub>2</sub> O <sub>4</sub>
		0.50	0.42	54.1	0.17	48	$\alpha$ -Fe <sub>2</sub> O <sub>3</sub>
<b>NiFe<sub>2</sub>O<sub>4</sub> coated with SiO<sub>2</sub> annealed</b>							
$T = 300$ K		0.26	0.83	-	0.29	100	NiFe <sub>2</sub> O <sub>4</sub>
$T = 77$ K		0.37	0.88	-	0.32	100	NiFe <sub>2</sub> O <sub>4</sub>