

s, p - d coupling in ZnO doped with 3d transition metal impurities

A. Ciechan, P. Bogusławski

Institute of Physics, Polish Academy of Sciences, al. Lotników 32/46, 02-668 Warsaw, Poland; e-mail: ciechan@ifpan.edu.pl

Motivations

- The $s, p - d$ coupling between free carriers and the localized d -electrons of the TM dopants constitutes the basic feature of diluted magnetic semiconductors. The magnetic properties of ZnO:TM are explored because of possible application in spintronics or solotronics [1].
- We study electronic and magnetic properties of ZnO:TM and determine dominant mechanisms of coupling with localized spins of magnetic atoms for both electrons and holes [2].

DFT methods

The calculations are performed within the DFT with the GGA+ U approximation [3]. We use the pseudopotential method implemented in the Quantum ESPRESSO code [4]. Analysis of a single TM impurity in ZnO is performed using $3 \times 3 \times 2$ supercells with 72 atoms.

U corrections for 3d(TM), 3d(Zn) and 2p(O) electrons:

	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	O
U (eV)	2.0	2.0	2.0	1.5	4.0	3.0	3.0	2.0	12.5	6.25

The values of U corrections give a correct band structure of ZnO and lead to an excellent agreement with experimental optical properties of TM ions in ZnO [5].

TM levels in ZnO

Effect of intrashell Coulomb repulsion [6]:
TM levels increase with increasing occupation

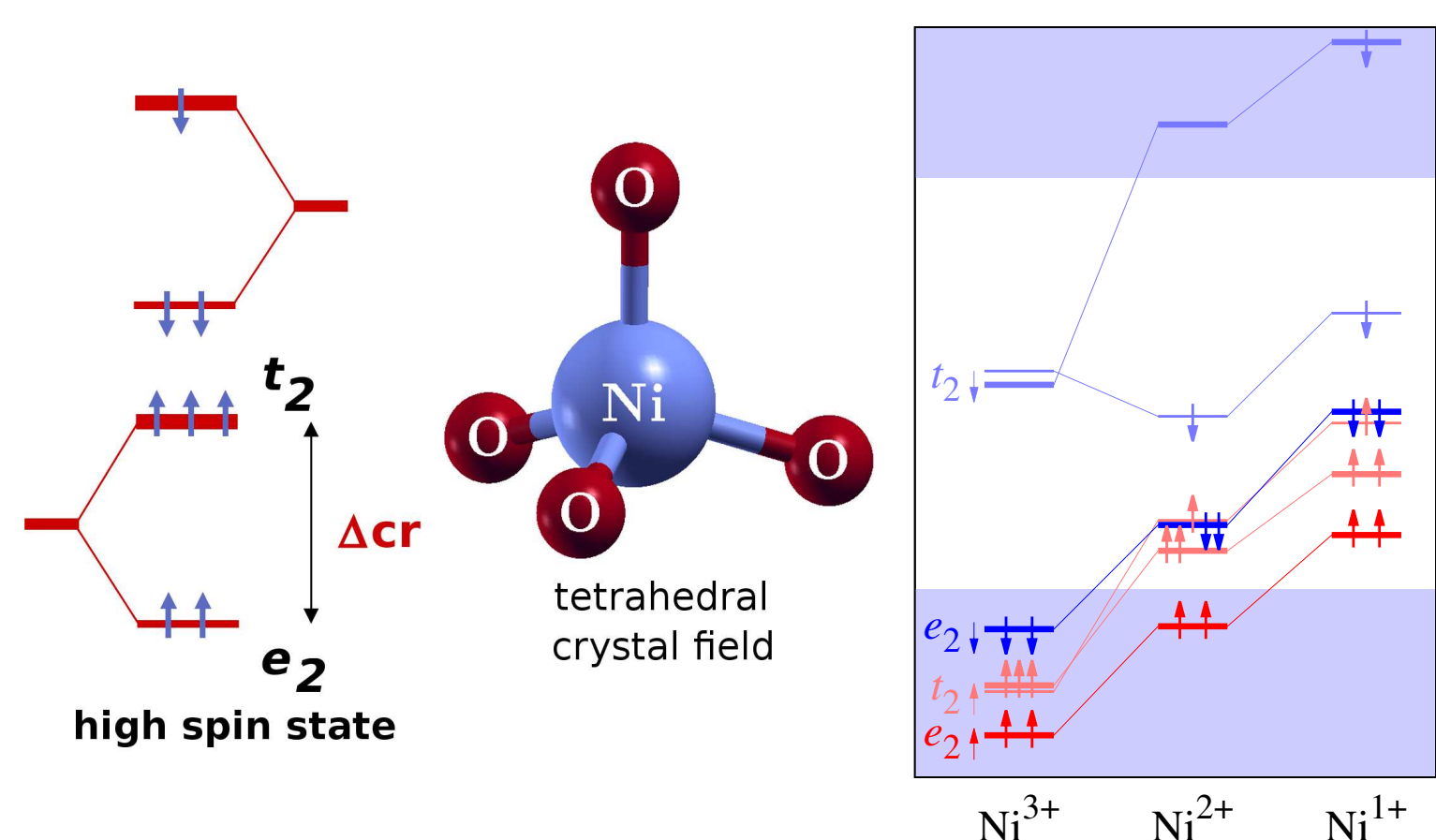


Figure 1: (left) Spin splitting and CF splitting of TM level in ZnO. (right) The energy levels of TM in various charge state.

TM²⁺ levels

levels decrease with increasing atomic number

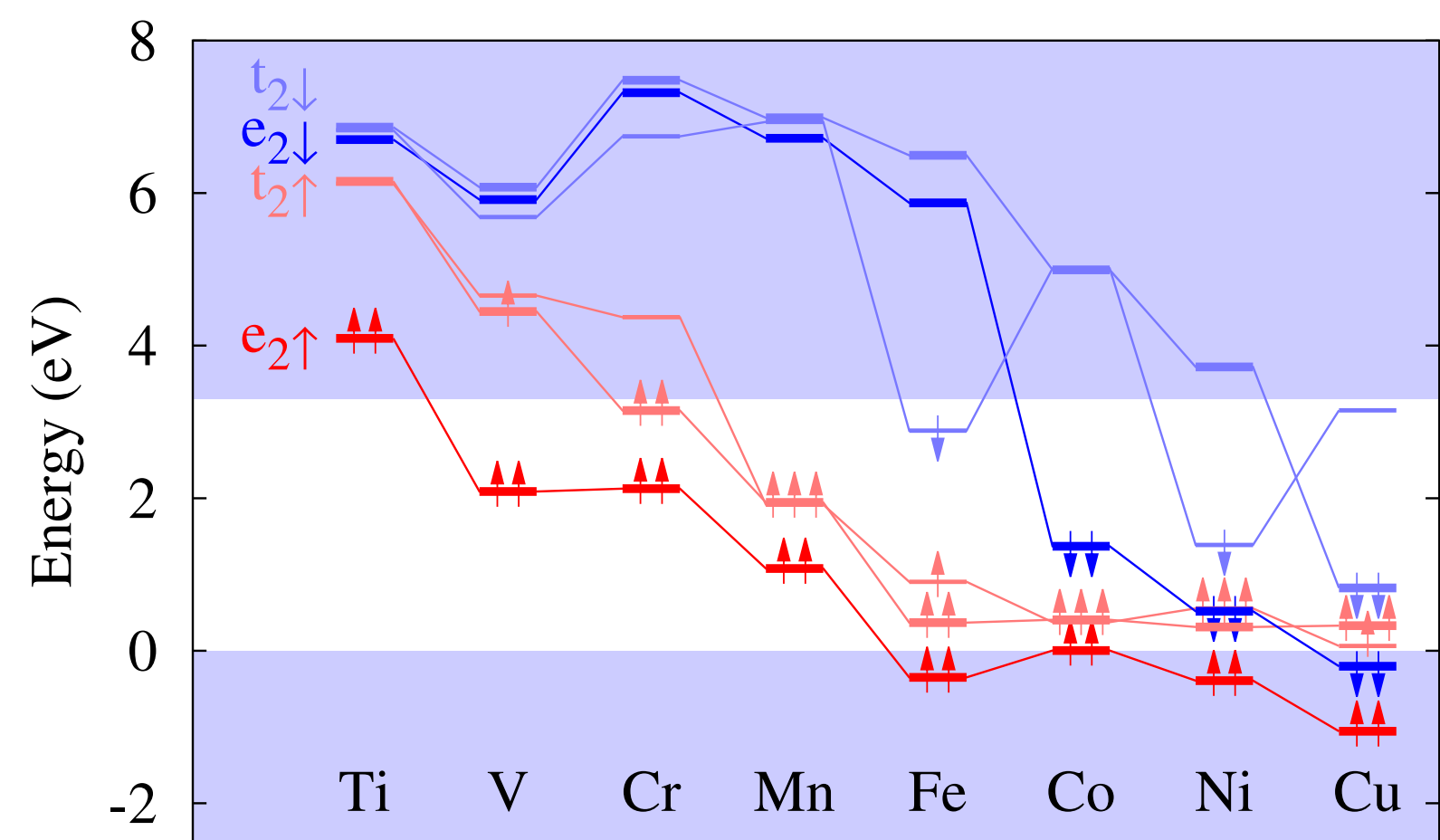


Figure 2: Energy levels of TM²⁺ ions in ZnO.

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sp-d magnetic coupling

Exchange constants are calculated from

$$N_0\alpha = \Delta\varepsilon_c / (x\langle S \rangle), \quad N_0\beta = \Delta\varepsilon_v / (x\langle S \rangle),$$

where $\Delta\varepsilon_c$ and $\Delta\varepsilon_v$ - spin splittings of CBM and VBM, x - concentration of the TM ions, $\langle S \rangle = M/2$ - total magnetic moment.

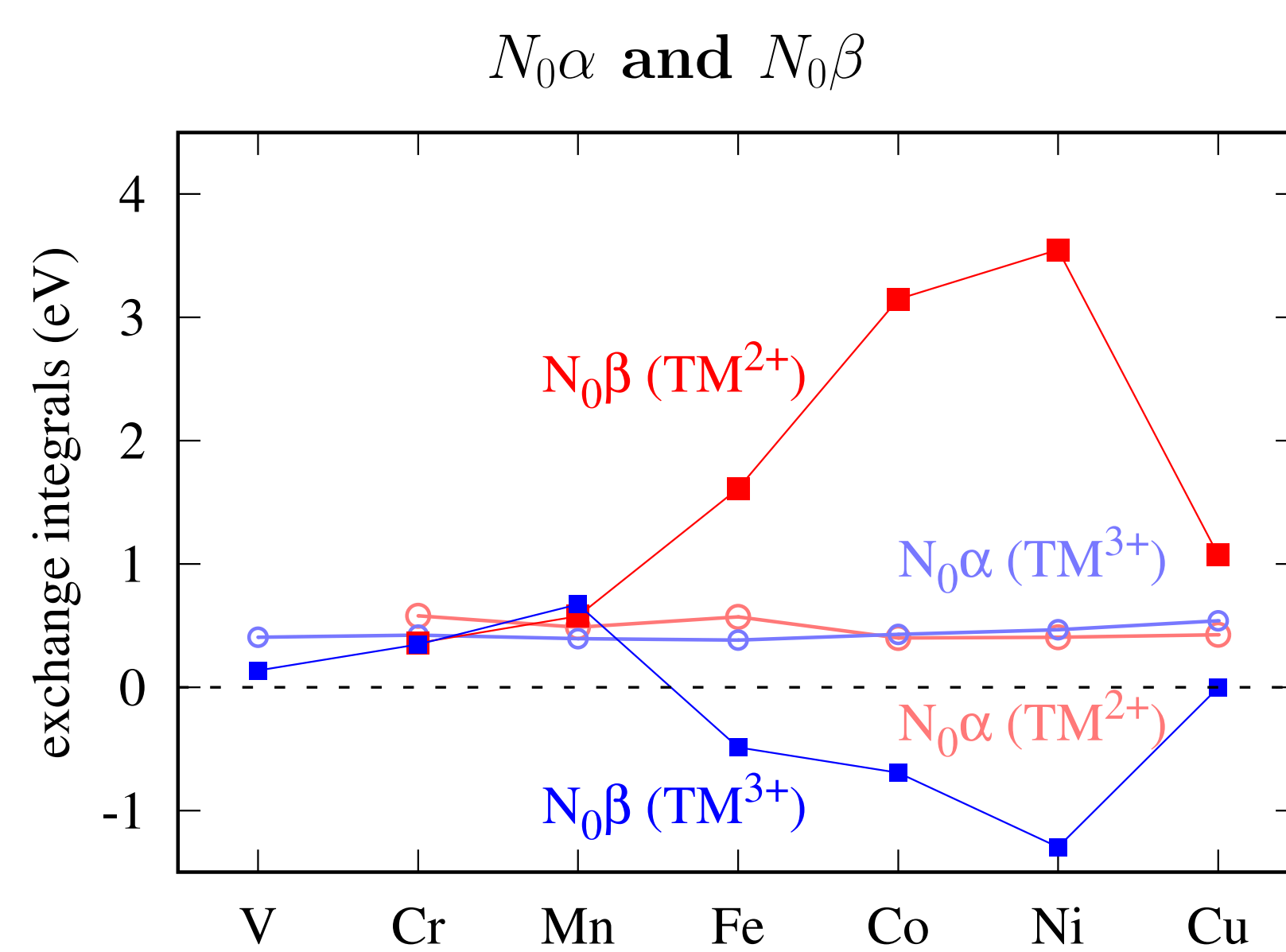


Figure 3: The exchange integrals $N_0\alpha$ and $N_0\beta$ for ZnO with TM in 2+ and 3+ charge state.

Spin density and xc perturbation

In spin polarized system $\Delta n = n_\uparrow - n_\downarrow \neq 0$ and thus

$$\Delta V_{xc} = V_{xc\downarrow} - V_{xc\uparrow} \approx A(n_\downarrow^{1/3} - n_\uparrow^{1/3})$$

$\Delta\varepsilon_c = \langle CBM \downarrow | V_{xc\downarrow} | CBM \downarrow \rangle - \langle CBM \uparrow | V_{xc\uparrow} | CBM \uparrow \rangle$
if $|CBM \uparrow\rangle = |CBM \downarrow\rangle$ then simply:

$$\Delta\varepsilon_c = \langle CBM | \Delta V_{xc} | CBM \rangle$$

Both Δn and ΔV_{xc} dominated by $d(\text{Co})$ with contribution of $p(\text{O-nn})$

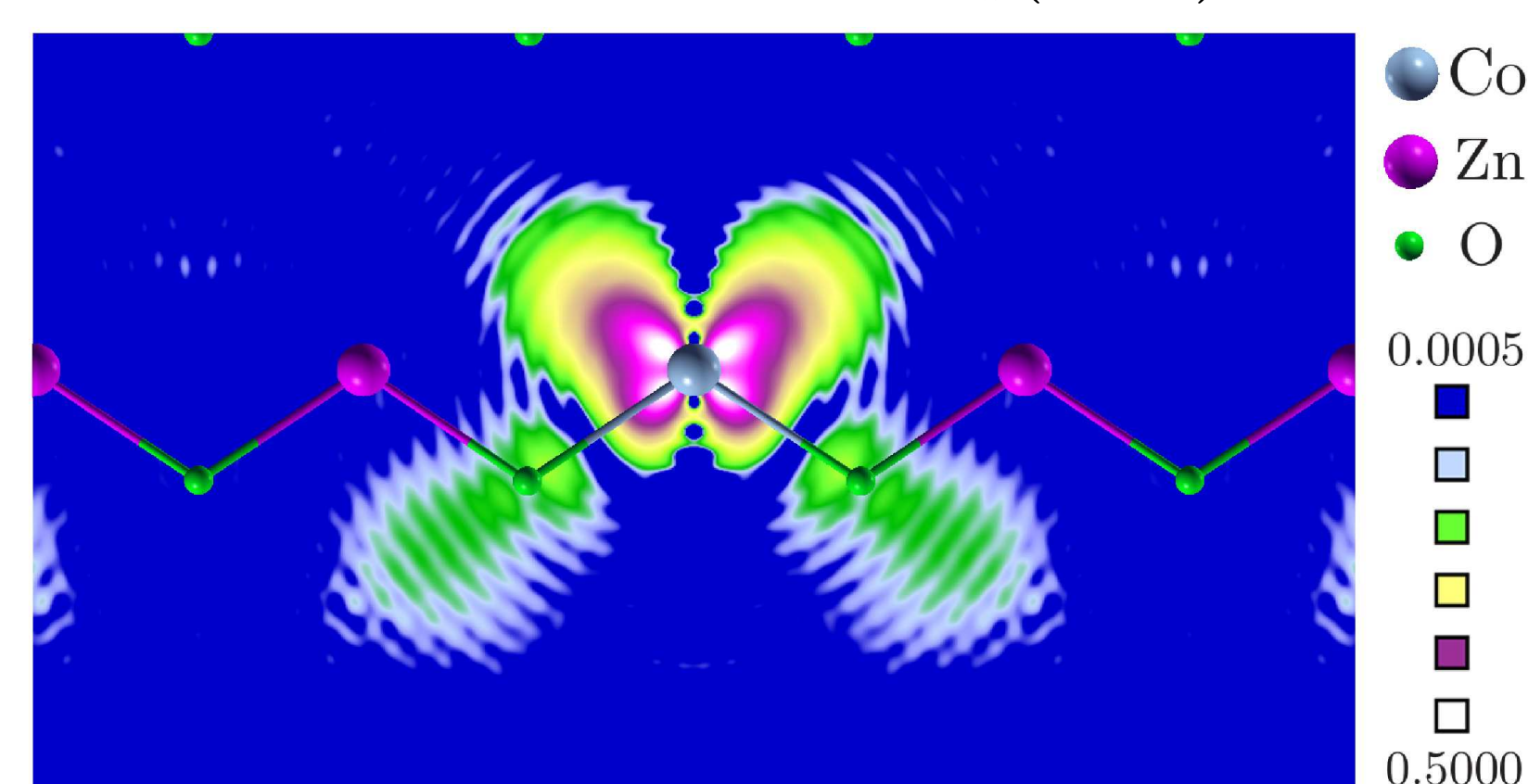


Figure 4: Two-dimensional plot of $n_\uparrow^{1/3} - n_\downarrow^{1/3}$ of ZnO with Co 2+ in the plane containing Co-O bonds.

N0alpha constant and s-d coupling

CBM: s(O)+s(Zn)+s(TM) states

- on-site direct exchange on TM ion (70%, Liu picture) and
- on O-nn caused by p(O)-d(TM) hybridization (30%, new effect)
- $s - d$ coupling is always FM, $N_0\alpha$ similar for all ions

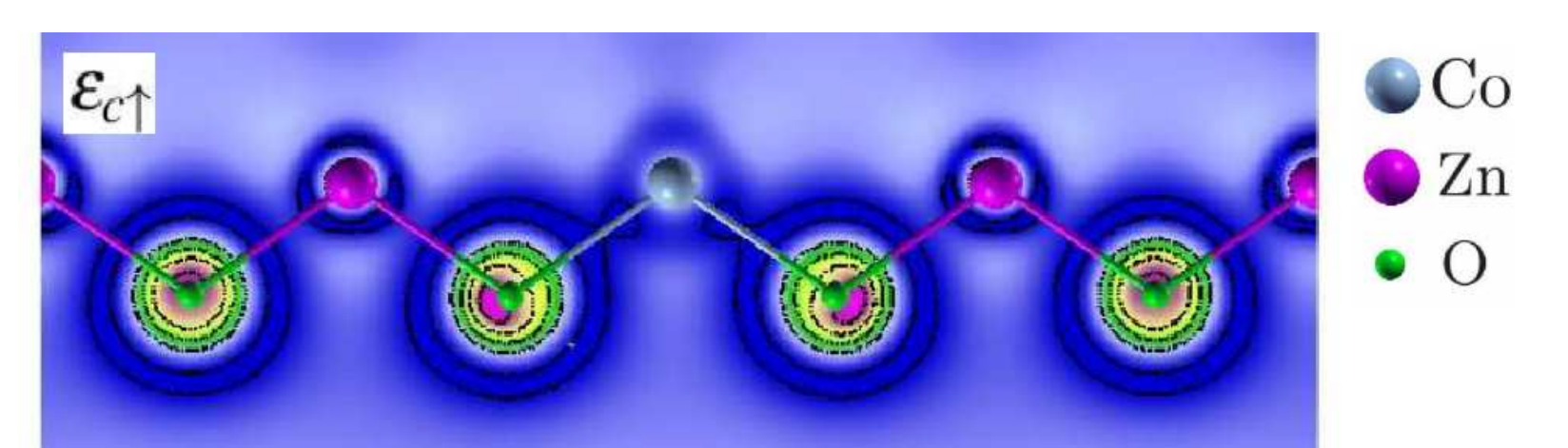


Figure 5: The wave functions squared of the CBM of ZnO:Co²⁺.

Comparison with experiment

Ti, V, Cr

experiment: $N_0\alpha = N_0\beta \approx 0$ [7]

theory: Ti - non-magnetic; V, Cr - very weak coupling

Mn - good agreement

experiment: $N_0|\beta - \alpha| = (0.2 \pm 0.1)$ eV [8]

theory: $N_0|\beta - \alpha| = 0.1$ eV for Mn²⁺

Co - close to experiment

experiment: $N_0|\beta - \alpha| = 0.8$ eV [9]

theory: $N_0|\beta - \alpha| = 2.6$ eV for Co²⁺

$N_0|\beta - \alpha| = 1.2$ eV for Co³⁺

N0beta constant and p-d coupling

VBM: p(O)+p(Zn)+d(TM)

- kinetic exchange caused by strong p(O)-d(TM) hybridization
- $sp - d$ can be FM ($N_0\beta > 0$) or AFM ($N_0\beta < 0$)

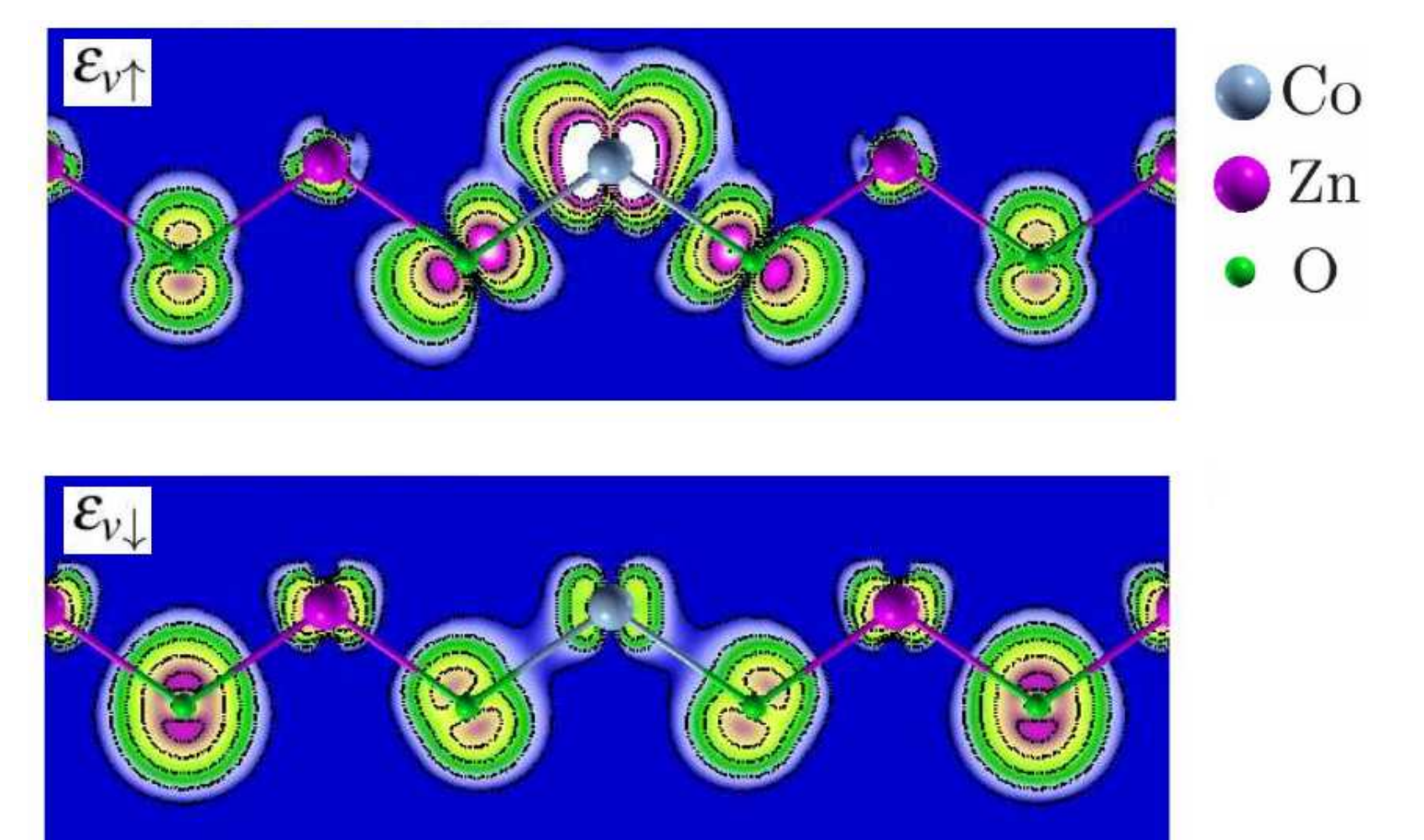


Figure 6: The wave functions squared of the VBM of ZnO:Co²⁺.

Sign of N0beta - Anderson model

$$N_0\beta \sim \Delta\varepsilon_v = \frac{1}{2} \left(\frac{|V_{hop}|^2}{\varepsilon(t_{2\uparrow}) - \varepsilon_v} - \frac{|V_{hop}|^2}{\varepsilon(t_{2\downarrow}) - \varepsilon_v} \right)$$

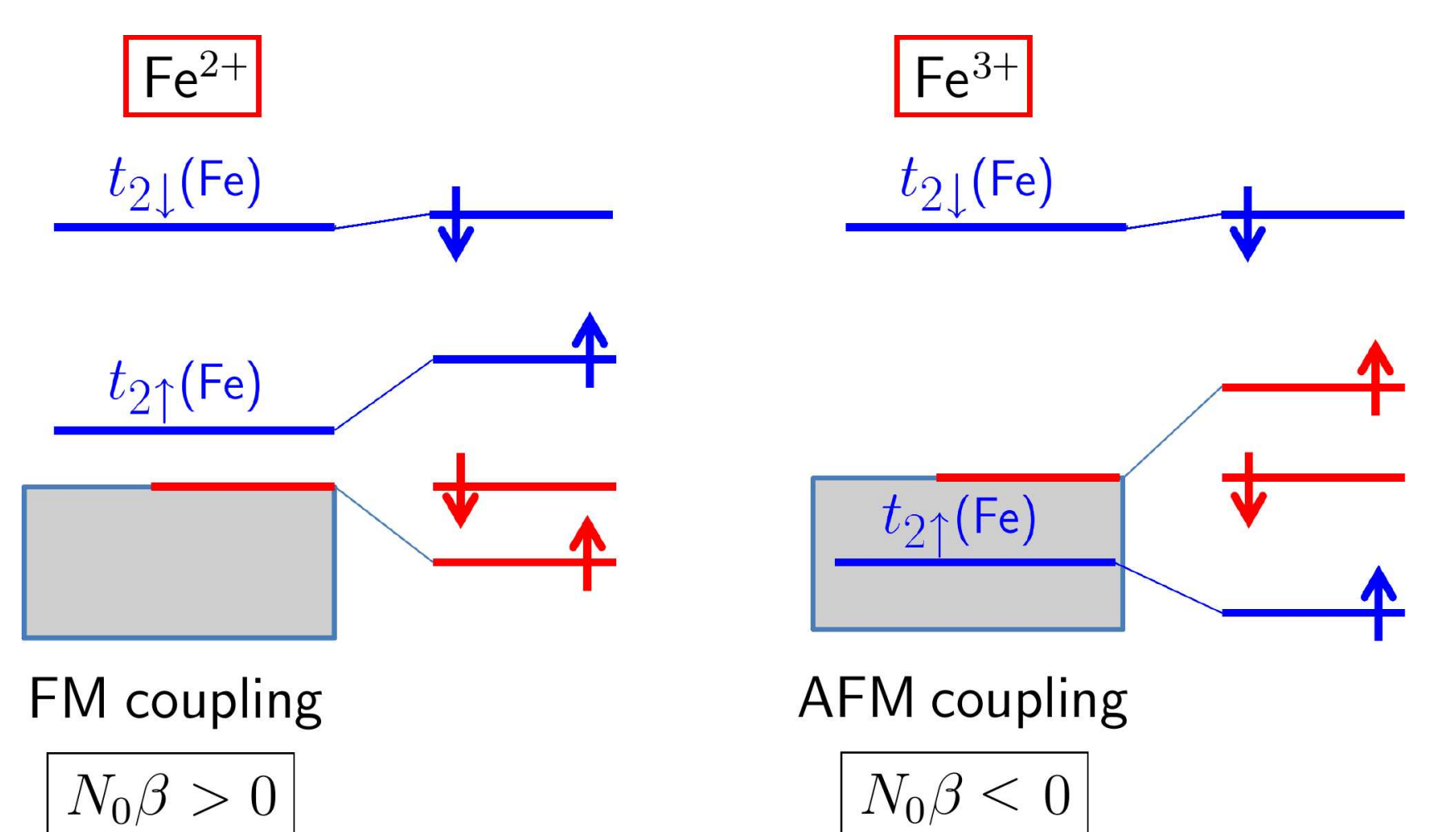


Figure 7: Dependence of the sign of $N_0\beta$ exchange constant on the position of t_{2g} levels of TM ion.

Change of sign of N0beta

dependence of $p - d$ coupling on the charge state caused by intra-shell Coulomb repulsion:

$N_0\beta < 0$ for Fe³⁺, Co³⁺, Ni³⁺, Cu³⁺

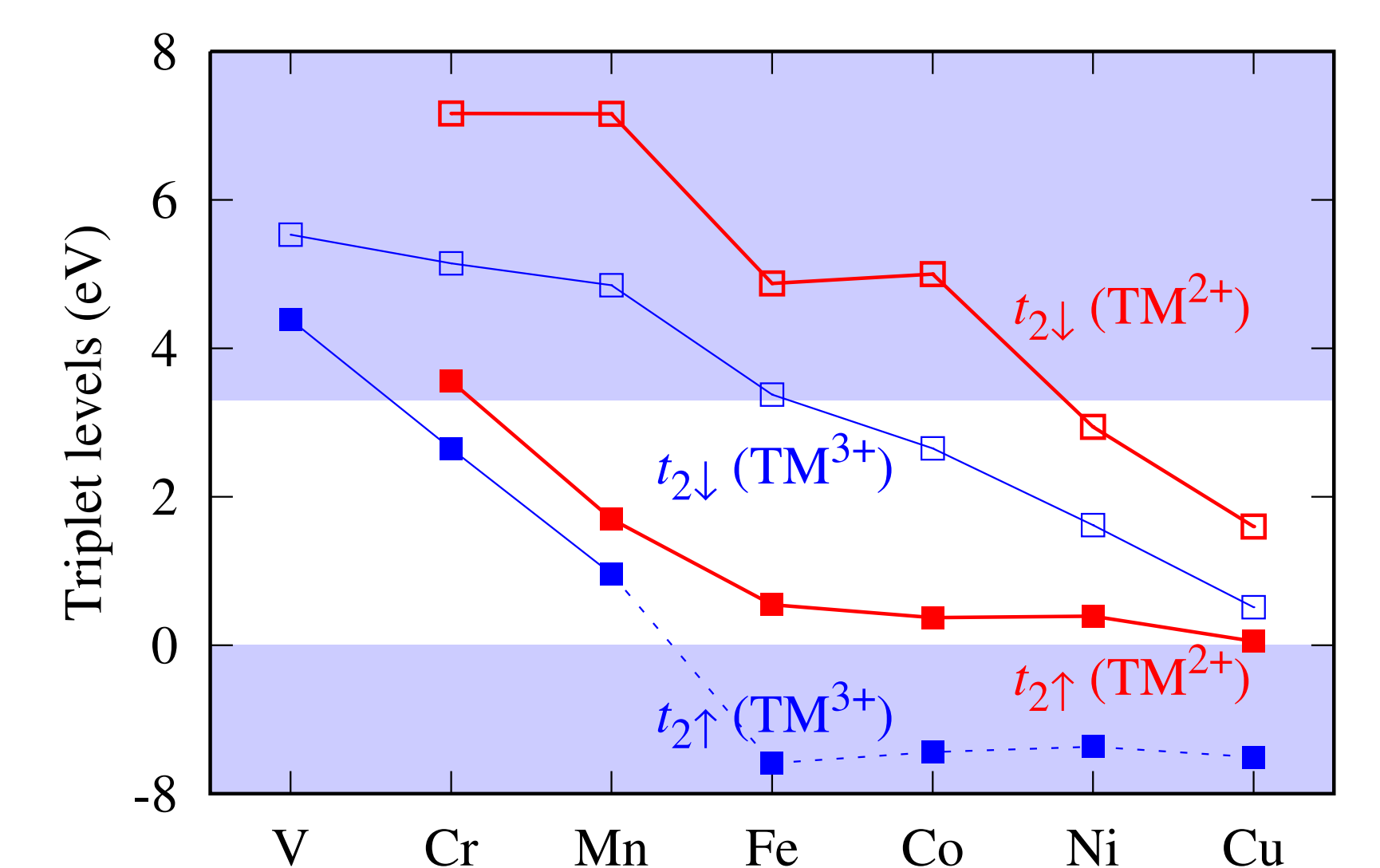


Figure 8: The triplet levels of TM²⁺ and TM³⁺ ions in ZnO.

Conclusions

- The $s - d$ coupling between magnetic moment of TM ion and conduction band electrons is always FM and the exchange constant $N_0\alpha \approx 0.5$ eV. It is caused by on-site direct exchange on TM ion as well as on nearest-neighbour O ions, which magnetic properties result from p(O)-d(TM) hybridization.
- In turn, the $p - d$ coupling with the valence band holes strongly depends on the position of TM levels in the gap. For some ions like Fe, Co, Ni, the coupling changes the sign from FM ($N_0\beta > 0$) to AFM ($N_0\beta < 0$) going from 2+ to 3+ charge state.

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