





# Molecular investigation of the coalescence dynamics of surfactant-laden droplets

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### Introduction

### **Key concepts:**

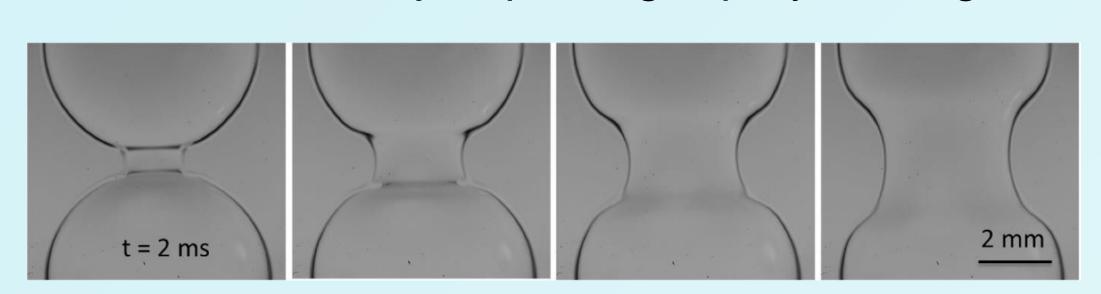
- ☐ Understand the coalescence dynamics of surfactant-laden droplets
- ☐ Optimize the coalescence process and inform surfactant design for relevant applications

### Aim:

☐ Reveal the mass transport mechanism and the role of key parameters in coalescence process

### **Applications:**

☐ Microfluidics, Inject printing, Spray cooling



Experimental image of droplets coalescence (aqueous solution SLES) [1]

## **Model and Methodology**

#### Model:

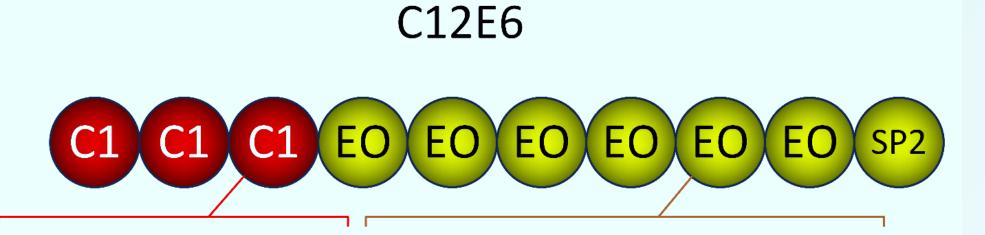
☐ Martini coarse grained force-field [2] Method:

■ Molecular dynamics simulation (NVT and NPT ensembles)

### **Materials:**

■ Water

■ Non-ionic surfactant C12E6



Hydrophobic

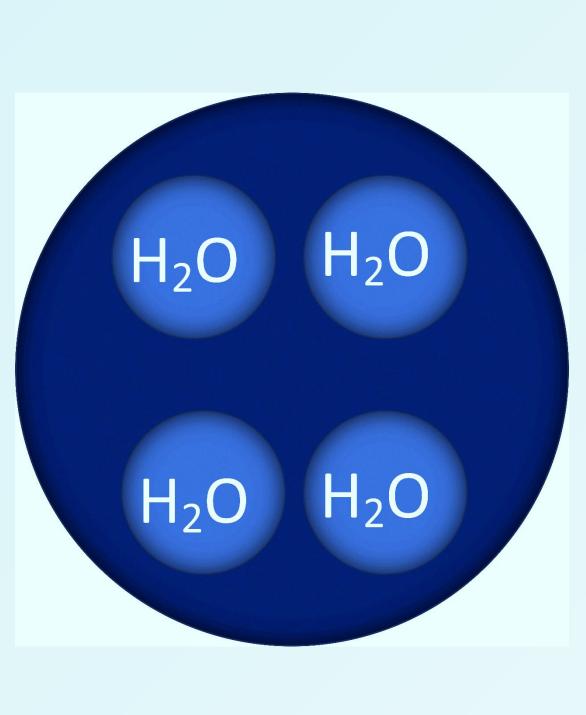
Hydrophilic

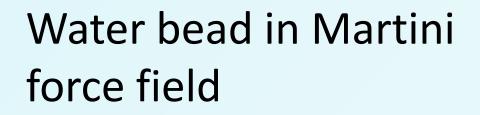
Martini representation of C12E6 surfactant

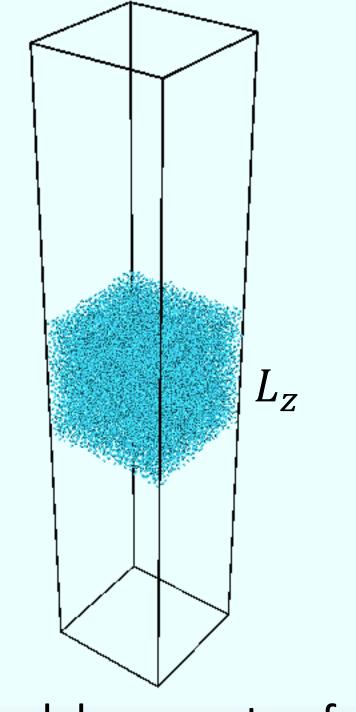
## **Model and Methodology**

Particle type (Martini)	Name
C1	Butane
EO	Ethylene oxide
SP2	Acetic acid

C12E6 beads definition in Martini force field [2]







Water slab geometry for surface tension calculation

## Results

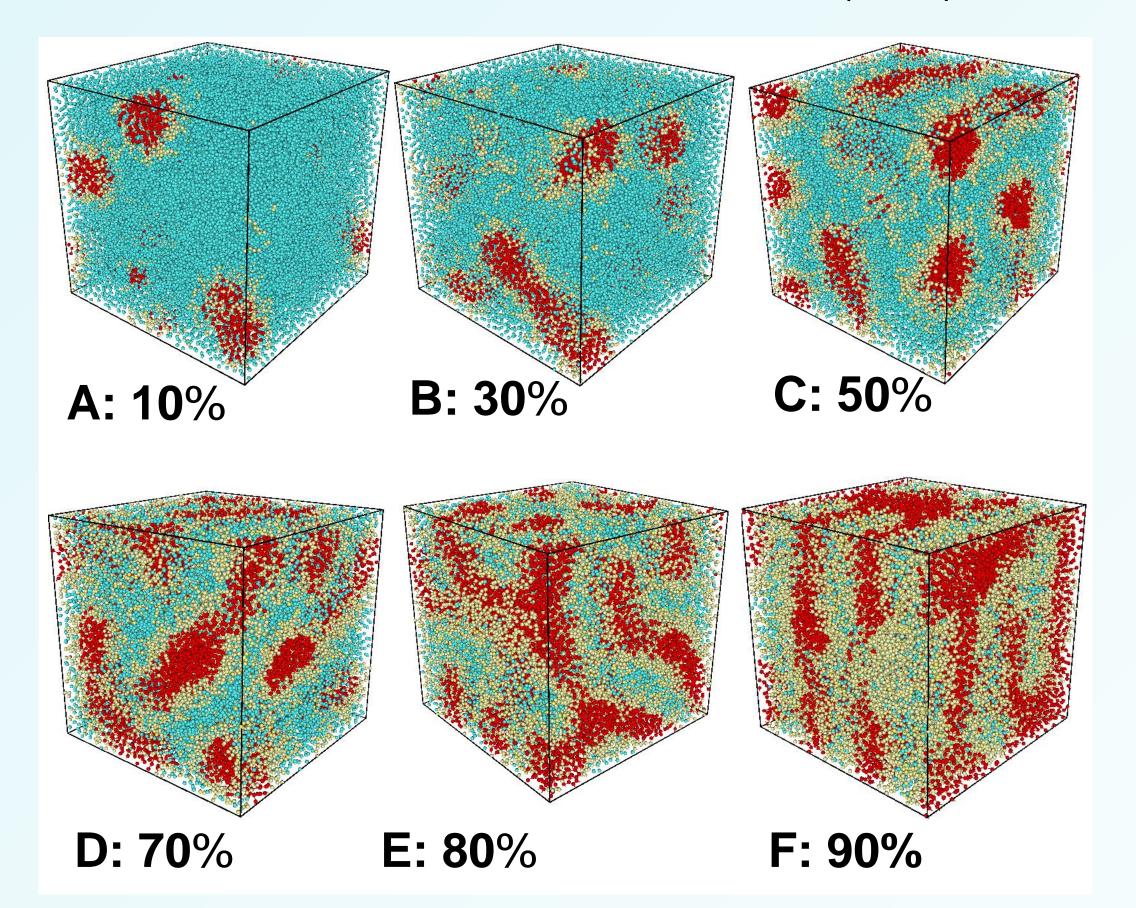
Evaluation of water surface tension (mechanical way), based on the relation:

$$\gamma = \frac{L_z}{2} \left[ P_{zz} - \left( \frac{P_{xx} + P_{yy}}{2} \right) \right]$$

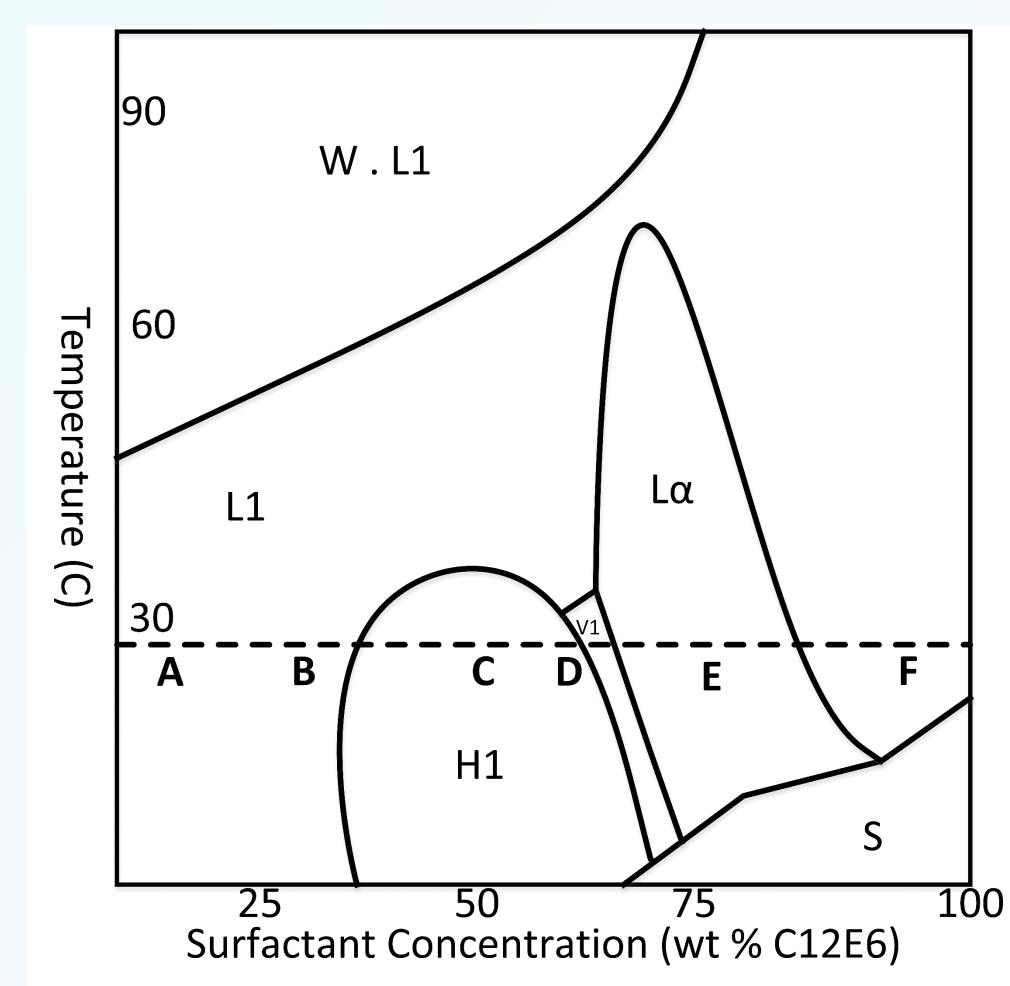
Martini version	cut-off [Å]	Surface tension [mN/m]
3	11	27.11
3	25	50.21
3	35	50.50
2	11	31.02
2	25	56.05
2	35	85.80
Surface tension of Water-air (25 °C): ~72 mN/m		

## Results

Morphologies of water-surfactant system at different concentrations of surfactant (wt%)



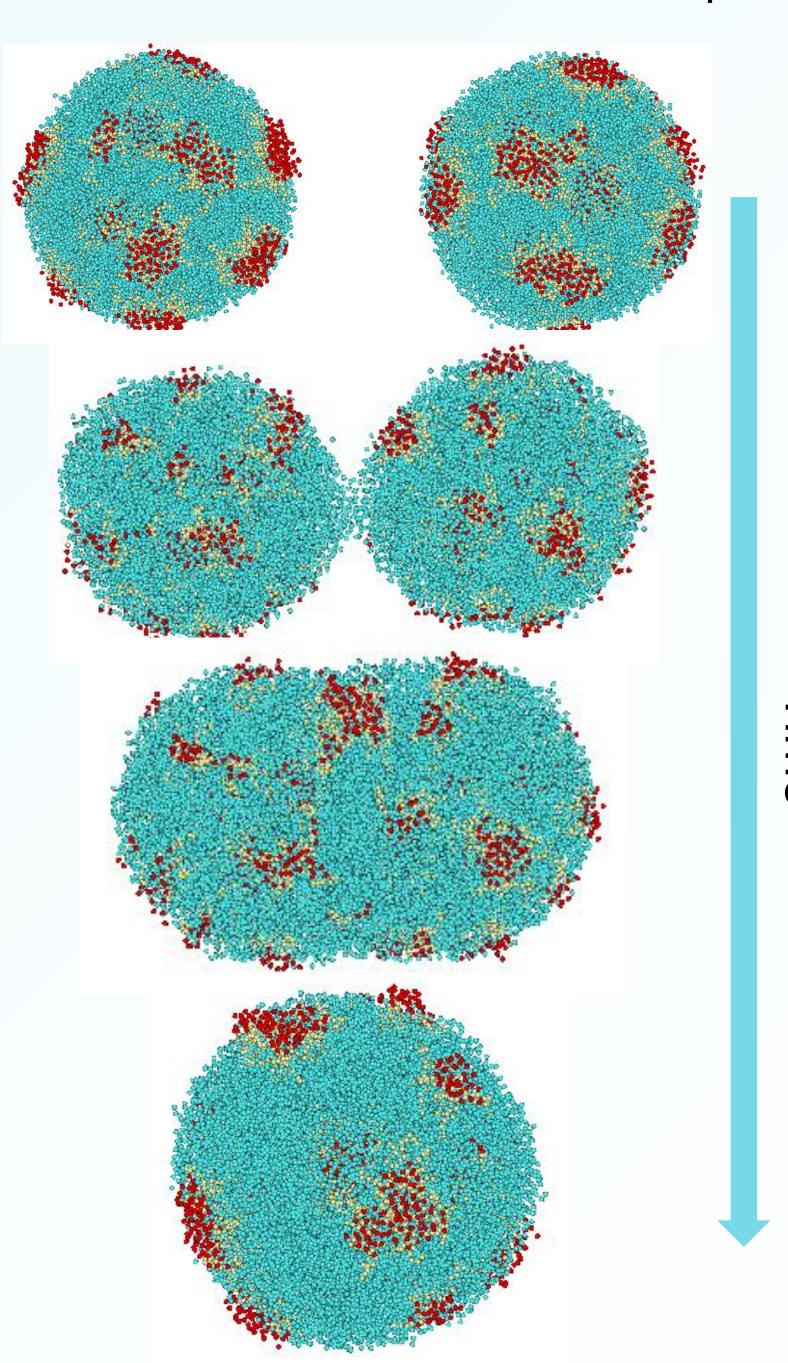
Experimental phase diagram of C12E6/water system [3]



Further simulations are needed to validate the phase behavior of the surfactant-water solution

## Results

Coalescence of surfactant-laden droplets



## Conclusion

- ☐ The Martini2.2 force-field with a cut-off of 25 Å indicates a close match with experimental results. Further model validation is required.
- ☐ We have presented early examples of droplet coalescence, which will enable us to carry out the full investigation of this phenomenon.

#### References:

- E. Nowak, N.M. Kovalchuk, Z. Che, M.J.H. Simmons. Colloids Surf. A. 505:124-131 (2016)
  S.J. Marrink, H.J. Risselada, S. Yefimov, D.P. Tieleman, A.H. de Vries. JPC-B, 111:7812-7824 (2007)
- 3. D. J. Mitchell, G. J. T. Tiddy, L. Waring, T. Bostock, M.P. McDonald, J. Chem. Solc. Faraday Trans. I 79:975-1000 (1983)

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